## **CAMBRIDGE INTERNATIONAL EXAMINATIONS**

**Cambridge International General Certificate of Secondary Education** 

## MARK SCHEME for the October/November 2015 series

## 0654 CO-ORDINATED SCIENCES

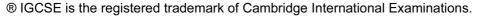
**0654/23** Paper 2 (Core Theory), maximum raw mark 120

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge will not enter into discussions about these mark schemes.

Cambridge is publishing the mark schemes for the October/November 2015 series for most Cambridge IGCSE<sup>®</sup>, Cambridge International A and AS Level components and some Cambridge O Level components.





Р	age 2	Mark Scheme	Syllabus	Paper
		Cambridge IGCSE – October/November 2015	0654	23
1	(a) (i)	fat ; protein ; calcium ;		[max 2]
	(ii)	iron ;		[1]
	(iii)	has more fat ;		[1]
	(b) (i)	(1.50) 15; 6;		[2]
	(ii)	no, because large amount is needed to meet vitamin C requirement	;	[1]
	(iii)	bleeding gums; poor skin/bruising; scurvy;		[max 2]
	(c) (i)	prevents constipation/promotes peristalsis;		[1]
	(ii)	(named) cereal grain/fruit/vegetable ;		[1]
				[Total: 11]
2	(a) (i)	idea of greater precision/accuracy;		[1]
	(ii)	neutralisation;		[1]
	(iii)	salt ; water ;		[2]
	(b) (i)	(first 35 cm³) decreased slowly/decreased from pH 13 to 12; (next 10 cm³) decreased rapidly/more quickly/decreased from pH1	2 to 2 ;	[2]
	(ii)	40 (cm <sup>3</sup> ); evidence of finding the volume at pH = 7;		[2]
	(iii)	take same amount/20.0 cm <sup>3</sup> of alkali; add 40 cm <sup>3</sup> of the acid <i>(allow ecf from (ii))</i> ;		[2]
	(iv)	white solid/solid sodium chloride;		[1]
				[Total: 11]

Syllabus

Paper

Page 3	Mark Scheme	Syllabus	Paper
	Cambridge IGCSE – October/November 2015	0654	23

3 (a) (i)

4

	(gai	mma)	X-rays	ultraviolet	(visible)	infra-red	(microwaves)	radio waves	
		(all fo	ur correct	2 marks, any	two correc	ct 1 mark) ;;			[2]
	(ii)	micro	waves;						[1]
(b)	(i)	label l	line at bas	e of fire/labe	I line where	e both rays	meet ;		[1]
, ,	(ii)		m) ± 1 mm			·			[1]
		e i		. ,.					
(C)	coll	ide witl	h walls of	in motion ; container ;				_	01
	torc	ce of co	ollisions ex	erts a pressu	re ;			Įma	ax 2]
(d)			penguin) ; area of foo						[2]
	(04.		ar 0	, , , , , , , , , , , , , , , , , , ,					[—]
(e)			,	) particles are even with co	•		ly arranged ;		[1]
								[Total	: 10]
									- 4 -
(a)	(i)	magn	esium + ca	arbon dioxide	→ magne	sium oxide	+ carbon ;		[1]
	(ii)		_	n of oxygen a ns oxygen an			al of oxygen;		
		_	_	oses oxygen				[ma	ax 2]
(b)	(i)	anode	e clearly la	belled;					[1]
	(ii)	chlori	ne;						[0]
		$Cl_2$ ;							[2]
(c)	(i)	carbo carbo	n ; n dioxide ;						[2]
	(ii)				of the pro	nduct/lead v	vill conduct electi	ricity :	[1]
	(")	iosi il	io di <del>d</del> uli ida	ai conductivity	, or the pro	auci, icau v	viii conduct electi	noity,	ניו

[Total: 9]

			Cambridge IGCSE – October/November 2015	0654	23
5	(a)	(i)	asexual;		[1]
		(ii)	no gametes/fertilisation involved; genetically identical;		[max 1]
	(b)	(i)	photosynthesis;		[1]
		(ii)	sexual reproduction ;		[1]
	(c)	(i)	anther/stamen;		[1]
		(ii)	sepal;		[1]
	(d)	bed	cause the fruits develop from the flowers ;		[1]
					[Total: 7]
6	(a)	(i)	crosses (X) marked on graph at 13–14s, 71s, 105s and 150s;		[1]
		(ii)	13–14 (s) ;		[1]
		(iii)	20(s);		[1]
		(iv)	C–D or G–H ; graph goes down ;		[2]
	(b)	(i)	thermal energy produces increased particle vibration; particle vibration is passed on from particle to particle; metals are good thermal conductors;		[max 2]
		(ii)	gas around filament heats up/gas expands; gas rises/gas less dense;		[2]
		(iii)	<pre>wavelength: distance between two waves; but distance between two peaks/two troughs/two identical points of waves; frequency: number of waves produced per second/number of wave fixed point per second;</pre>		
	(c)	(i)	(current) = $\frac{\text{voltage}}{\text{resistance}}$ ; = $\frac{12}{4}$ = 3 (A);		[2]
		(ii)	8 (Ω) ;		[1]
				I	Total: 15]

Page 4

Syllabus

Paper

Page :	5	Mark Scheme	Syllabus	Paper
7 (0)		Cambridge IGCSE – October/November 2015	0654	23
7 (a)	Χy	rlem ;		[1]
(b)	fro fo	vaporation of <u>water</u> ; om surfaces of mesophyll cells; llowed by loss of <u>wate</u> r vapour; v diffusion;		
	Ol	ut through stomata ;		[max 4]
(c)	(i	(coloured) water does not move as far ;		[1]
				[Total: 6]
8 (a)		etroleum ; actional distillation ;		[2]
(b)	(i	carbon dioxide ; water ;		[2]
	(ii	reference to carbon monoxide/incomplete combustion; which are toxic/which could poison people;		[2]
(c)	(i	hydrocarbon will decolourise bromine ; if it is unsaturated ;		[2]
	(ii	) H C=C		
		H ; carbon – carbon double bond ; 4 × H <b>and</b> all else correct ;		[2]
				[Total: 10]
9 (a)	no	resultant force because constant speed ;		[1]
(b)	h	ree straight lines ; orizontal lines from boat and into eye ; ternal reflection shown at both prisms ;		[3]
(c)	(i	lead/concrete/aluminium ;		[1]
	(ii	geiger counter/GM tube, etc.;		[1]
				[Total: 6]

Page 6		6	Mark Scheme Cambridge IGCSE – October/November 2015		Paper 23
40	(-)	(1)		0654	
10	(a)	(ii)	deforestation; logging; building of roads/towns/factories; farming; fuel;		[1] [max 2]
	(b)	(ca	trol of hunting/nature reserve/conservation area; ptive) breeding programmes; rnatives to timber/control of deforestation/replanting; ;		[max 2]
	(c)	(i)	grow/photosynthesise more (because not eaten by okapis);		[1]
		(ii)	have less food/must find alternative food sources; (accept: more competition for food/migration)		[1]
					[Total: 7]
11	(a)	(i)	neon;		[1]
		(ii)	proton/atomic number/number of electrons;		[1]
		(iii)	9 protons ; 10 neutrons ;		[2]
	(b)	(i)	sodium chloride ;		[1]
		(ii)	reference to loss of electron(s)/loss of outer shell;		[1]
		(iii)	balance of charge/protons and electrons in the atom; excess of electrons in the ion/gains electrons;		[2]
	(c)		er nitrate ; te precipitate ;		[2]
					[Total: 10]
12	(a)	(i)	5000000(N);		[1]
		(ii)	need positive resultant, for upward motion/acceleration;		[1]
		(iii)	chemical, thermal (heat), kinetic ;; (all three for 2 marks, any two for 1 mark)		[2]

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	Cambridge IGCSE – October/November 2015 0	654	23
(b) (i)	sound waves cannot travel through space/vacuum or sound waves neemedium;	ed a	[1]
(ii)	time		
	$=\frac{225000000}{750}=300000(\text{km/s});$		[2]
(c) (i)	ionising radiation that humans are exposed to/radiation that is always t	:here ;	[1]
(ii)	rocks;		[1]
			[Total: 9]
13 (a) (i)	increased rate of breathing; increased depth of breathing/volume of breaths;		[2]
(ii)	more carbon dioxide/CO <sub>2</sub> ;		
	more water vapour ; warmer ;		[max 2]
(b) (i)	increased heart/pulse rate ; increased blood glucose ; AVP ;		[max 2]
(ii)	chemical/substance produced by a gland; carried in the blood;		

alters the activity of target organ(s);

destroyed by the liver;

**Mark Scheme** 

**Syllabus** 

**Paper** 

[max 3]

[Total: 9]