

## **Cambridge International Examinations**

Cambridge International General Certificate of Secondary Education

## **CO-ORDINATED SCIENCES**

0654/63

Paper 6 Alternative to Practical

October/November 2016

MARK SCHEME
Maximum Mark: 60

## **Published**

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Page 2	Mark Scheme	Syllabus	Paper
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Question			Ans	wers	Marks
1(a)	Nutrient tested for	Testing solution	Heat needed? (Yes/no)		3
	Protein	Biuret	no		
	Reducing sugar	Benedict's	yes		
	Starch	lodine	no		
	3 correct reagen	ts = 2, 1 correct = 1;	;		
	1 mark for heat f	or reducing sugar <b>or</b>	nly;		
1(b)	Testing solution used	n Initial colour	Colour after test		2
	Benedict's solut	ion Blue	blue		
	biuret solution	n Blue	purple/lilac		
	iodine	brown	brown		
	purple/lilac for p negative colours	rotein ; brown and blue ;			
1(c)	Benedict's: yellov	w/green/orange/re	d ;		2
	iodine: blue-blac	<b>k</b> ;			
1(d)	same volume of	juice and lemonade Benedict's solution ; small amount of red		e/red for high(er) amount of reducing sugar ;	3
				Total:	10

Page 3	Mark Scheme	Syllabus	Paper
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Question	Answers	Marks
2(a)	30 ; 32 ;	2
2(b)(i)	31 ;	1
2(b)(ii)	0.032 0.019/0.018, 0.012/0.011;	1
2(c)(i)	linear scale for vertical axis using at least half of the grid; all three points plotted correctly to within half a small square; best appropriate straight line or curve through the origin;	3
2(c)(ii)	as concentration increases speed increases ;	1
2(d)	0.75 and difference between them is much greater than difference between other pairs/% difference greater than other pairs/% difference greater than 10%;	1
2(e)	(reacted chips have) smaller surface area/(already reacted chips will) react slower;	1
	Total:	10

Page 4	Mark Scheme	Syllabus	Paper
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Question	Answers	Marks
3(a)(i)	correct symbol for voltmeter ; correct parallel (voltmeter) connection between X and Y ;	2
3(a)(ii)	1.9 (V);	1
3(a)(iii)	0.24 (A);	1
3(a)(iv)	7.9 ; Unit $\Omega$ /ohm ;	2
3(b)	<b>15</b> (Ω);	1
3(c)	YES (no mark) and values of $R_T$ and $0.5R_S$ are close enough / difference can be attributed to experimental error ;	1
3(d)	resistors become hot/temperature affects resistance ;	1
3(e)	increases;	1
	Total:	10

Page 5	Mark Scheme	Syllabus	Paper
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Question				Ans	wers	Marks
4(a)	Test- tube	Initial colour	Final Colour	Change in CO2 concentration (increase/decrease/no change)		1
	Α	red	purple	decrease		
	В	red	yellow	increase		
	С	red	red	(no change)		
	D	red	red	no change		
		for A <u>and</u> E other than	3 correct ; no change for [	O = no marks		
4(b)(i)	photosyn	nthesis (ren	noves CO <sub>2</sub> ) ;			1
4(b)(ii)	respiration	on (produce	es CO <sub>2</sub> ) ;			1
4(b)(iii)	rate of pl	notosynthe	sis and respirati	on is matched ;		1
4(c)	control/t	o show no	change without	organisms ;		1
4(d)(i)	water ba between	th ; 10–40 <u>°C</u> ;				2
4(d)(ii)	size of po amount of light inter type of w	size of tadp ondweed; of indicator nsity;				max 3
					Total:	10

Page 6	Mark Scheme	Syllabus	Paper
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Question	Answers	Marks
5(a)(i)	symbol for cell / DC power supply ; symbol for ammeter or lamp ;	2
5(a)(ii)	copper oxide / CuO ;	1
5(b)(i)	thermometer and stopper; thermometer bulb opposite to the side-arm;	2
5(b)(ii)	99.5 (°C);	1
5(b)(iii)	(0.5 less than 100) within experimental error/inaccuracy of thermometer/height above sea level;	1
5(c)	carbon dioxide / CO <sub>2</sub> ;	1
5(d)	sodium hydroxide solution / add ammonia solution ; colour of ppt. / specific example, e.g. blue ppt. = Cu <sup>2+</sup> ;	2
	Total:	10

Question	Answers	Marks
6(a)(i)	29 <u>.0</u> ; 41 <u>.0</u> ;	2
6(a)(ii)	eye level/bottom of meniscus;	1
6(b)	1.2 (1.193103448275862); 0.8 (0.83902439024463);	2
6(c)(i)	(teat/dropping) pipette;	1
6(c)(ii)	formula takes it into account ;	1

Page 7	Mark Scheme	Syllabus	Paper
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Question	Answers	Marks
6(c)(iii)	find the average/mean (of the three results for each liquid);	1
6(d)(i)	S and because it is less dense than water/liquid T;	1
6(d)(ii)	oil/S on top (ecf) with <b>one</b> line at 20 ;  water and Salt Solution  (water and salt solution or water and solution <b>R</b> )	1
	Total:	10