UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS International General Certificate of Secondary Education

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0654 CO-ORDINATED SCIENCES

0654/03

Paper 3 (Extended Theory), maximum raw mark 100

This mark scheme is published as an aid to teachers and students, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began.

All Examiners are instructed that alternative correct answers and unexpected approaches in candidates' scripts must be given marks that fairly reflect the relevant knowledge and skills demonstrated.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

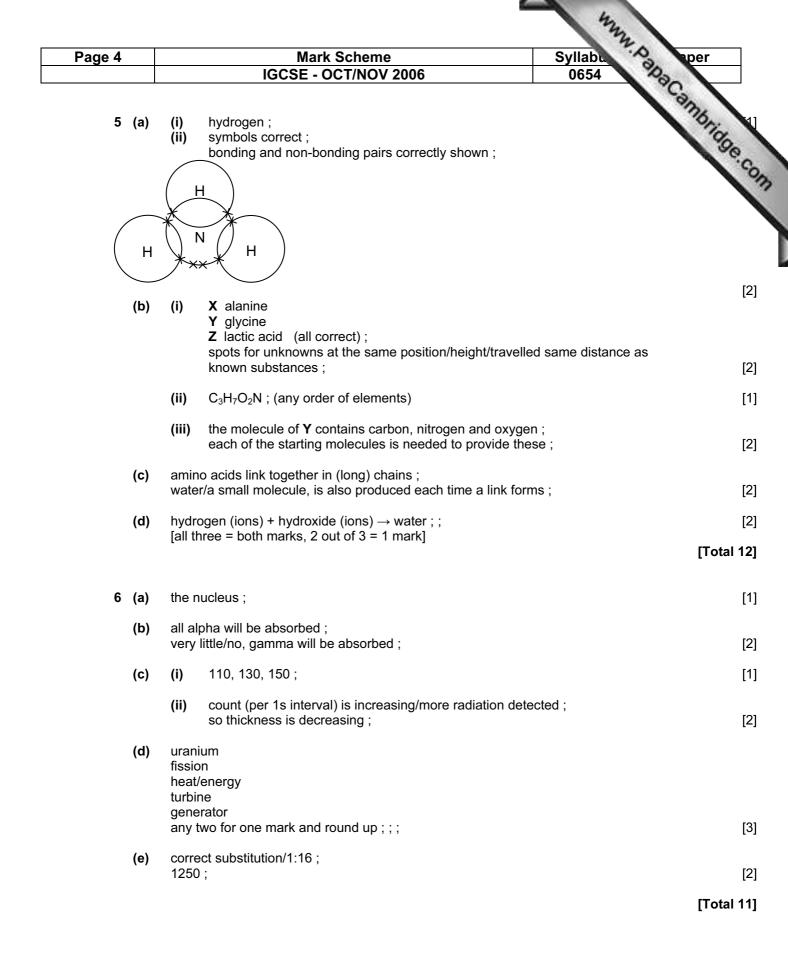
The grade thresholds for various grades are published in the report on the examination for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level syllabuses.

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CIE is publishing the mark schemes for the October/November 2006 question papers for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level syllabuses and some Ordinary Level syllabuses.

Page 2	Mark Scheme Syllab.	per
	IGCSE - OCT/NOV 2006 0654	
1 (a)	Mark Scheme Syllability IGCSE - OCT/NOV 2006 0654 key is made up of pairs of genuine choices ; each choice could be made if only one organism in front of you (i.e. not just 'long beak. short beak) ; completed in no more than four pairs ; all birds key out correctly ;	ambridge.c.
(b)	it is its Latin name ; first word is its genus ; second word is its species ;	[2
(c)	idea that some birds might have features not possessed by others ; birds that could begin to fly have better chance of survival ; because they can escape predators ; these birds most likely to reproduce ; and pass their genes onto offspring ;	[max 4
		[Total 10
2 (a)	<u>coal</u> <u>methane</u> ;	[1
(b)	(i) M_r of heptane = $(12 \times 7) + (16 \times 1) = 100$; moles of heptane = $684 \div 100 = 6.84$;	[2
	(ii) (car uses 6.84 moles of heptane) moles of carbon dioxide = $6.84 \times 7 = 47.88$; M_r carbon dioxide = $(12 \times 1) + (16 \times 2) = 44$; mass of carbon dioxide = $47.88 \times 44 = 2106.7g/2.1kg$;	[3
	(iii) gasoline is not just heptane/combustion is not complete/data is only an average/actual will depend on driving conditions ;	[1
(c)	 (i) magnesium and copper ; magnesium and copper have greatest reactivity difference ; voltage/p.d./is greater the greater the difference in reactivity ; 	[3
	 (ii) car battery is recharged (when engine working)/torch battery not recharged ; chemicals in car battery are not used up/torch batteries contain chemicals which are used up ; 	[2

per	Syllabu	Mark Scheme	Page 3
	0654 230	IGCSE - OCT/NOV 2006	
mbridge	Syllabt 0654 is volume of object ;	 (i) add object to known volume of water ; volume of water displaced/difference in volumes 	3 (a)
.9		 density = mass/volume ; 0.25 kg/dm³ ; accept other correct units 	
			(b)
[2		= 40 000 J ; (4 x 10 x 1000)	
-			(c)
		by friction ;	
[3		from man/clothing, to tent ; <i>accept other way round</i>	
	aster'	 Sun/heat, causes particles to move faster ; not 'vibrate some particles will be moving faster than others ; 	(d)
	come intermolecular forces ;	fastest particles have enough energy, to escape/to over	
[max 3		wind carries away water particles ;	
[Total 12			
[2		 A scapula /shoulder blade B ulna C humerus D tendon any two correct for one mark ; ; 	4 (a)
-			(b)
		triceps contracts ;	(0)
[3		triceps, gets shorter/pulls on B /pulls on ulna ;	
[1		(i) line to space within elbow joint or shoulder joint	(c)
[1		(ii) lubrication/reduce friction ;	
[1) (i) in central nervous system/in brain/in spinal cord	(d)
		(ii) long axon ; carries impulse quickly ;	
		fatty sheath ; insulates/speeds impulse ;	
		(many) dendrites/synapses, on cell body ;	
[max 3		receive impulses from other neurones ;	



Page 5	Mark Scheme Syllabu	per
	IGCSE - OCT/NOV 2006 0654	Pac
7 (a)	has a cell wall ; has a (large) vacuole ;	apacampringe.co
(b)	(i) lime water ; goes milky ;	9. CC
	 (ii) respiration/fermentation ; by yeast/fungus (cells) ; glucose combining with oxygen ; 	[max 2]
(c)	(i) B and C ;	[1]
	(ii) D and E ;	[1]
	(iii) mark between 4.4 hours and 6 hours ;	[1]
	(iv) (shortage of) glucose/oxygen ;	[1]
	 (v) add more glucose/add more oxygen ; if number of yeast cells increases then this was a limiting factor ; 	[2]
		[Total 12]
8 (a)	rusting not expected in either tube ; rusting requires, air/oxygen, <u>and</u> water (together) ; nail in A has no water ; nail in B has no, air/oxygen ;	[max 3]
(b)	one of the products is an alloy ; alloys are formed by mixing molten metals ; high temperatures required to obtain molten metals ;	[max 2]
(c)	(i) Cr ³⁺ ; reference to balancing of charges e.g. 2 x 3+ and 3 x 2- ;	[2]
	(ii) chromium oxide + sulphuric acid \rightarrow chromium sulphate + water ;	[1]
	(iii) (negative charge needed) to attract positive chromium ions;	[1]

