CAMBRIDGE INTERNATIONAL EXAMINATIONS

International General Certificate of Secondary Education

MARK SCHEME for the May/June 2013 series

0654 CO-ORDINATED SCIENCES

0654/23 Paper 2 (Core Theory), maximum raw mark 120

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge will not enter into discussions about these mark schemes.

Cambridge is publishing the mark schemes for the May/June 2013 series for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level components and some Ordinary Level components.



Page 2	Mark Scheme	Syllabus	Paper
	IGCSE – May/June 2013	0654	23
A labelle	ed at either 0s or 50s; ed between 0s and 20s or between 40s and 50s; ed between 20s and 40s;		[3]
goe son	ximum of 2.8 at 1.00 p.m.; es up then down; ne energy for 17 hours; er use of data;		[max 2]
(ii) ene	ergy input from Sun varies ;		[1]
to releas heat tur steam d	fossil fuel is burned ; se <u>heat</u> energy ; ns water into steam ; Irives turbine ; drives generator ;		[max 3]
• •	er in parallel with photocell ; symbol ;		[2]
(e) lateral ir upright	nversion ; ;		[2] [Total : 13]
gas dies	nery gas – bottled gas/camping gas/other correct soline – car engine fuel/fuel for petrol engines; sel – fuel for diesel engines; bon dioxide;	·,	[3]
wat			[2]
(iii) hyd	Irocarbons react with oxygen/oxygen bonds to the h	ydrocarbon ;	[1]
(b) (i) (cat	talytic) cracking ;		[1]
B r	mixture remains orange; mixture changes from orange to colourless; ne unsaturated hydrocarbons formed (during crackin saturated hydrocarbons react with bromine/decoloris		[4]

1

2

[Total: 11]

Page 3	Mark Scheme	Syllabus	Paper
	IGCSE – May/June 2013	0654	23

3 (a)

producer	consumer	herbivore	carnivore
bilberry and spruce ;	any two of: vole, Siberian jay, red squirrel, lynx, goshawk;	any two of: vole, Siberian jay, red squirrel ;	lynx and goshawk ;

[4]

(b) loss of soil/soil erosion; flooding;

carbon dioxide build up ; [max 2]

(c) muscle contraction; protein synthesis; cell division; passage of nerve impulses; maintenance of constant body

maintenance of constant body temperature;

digestion; [max 3]

[Total: 9]

4 (a) (i) gas produced;

temperature change; [2]

(ii) hydrogen; lighted splint pops;

[2]

(iii) solution is becoming less acidic;

because dilute hydrochloric acid is reacting/being used up;

[2]

(b) (i) the volume of gas trapped in the measuring cylinder;

the time taken for this volume to collect;

[2]

(ii) concentration of acid;

temperature of acid; surface area of magnesium;

[max 2]

[Total: 10]

	Page 4 Mark Scheme Syllabus		Paper			
			_	IGCSE – May/June 2013	0654	23
5	(a)	.,		-red;		[1]
		(11)	wave	elength/frequency;		[1]
	(b)	B hi	its at	sing into a more dense medium ; an angle greater than the critical angle ; sing into a less dense medium ;		[3]
	(c)	(i)	2 pro	otons and 2 neutrons/helium nucleus ;		[1]
		(ii)	rock	s etc ;		[1]
		(11)	TOCK	5 GIO ,		ניו
	(d)			rticles cannot pass through a thin sheet of lead/alp a few centimetres of air ;	ha particles can	pass [1]
	(e)			lüller tube ; phic film ;		[2]
	(f)	alph ionis can	na rac satior cer o	na radiation and beta radiation pass easily through to diation damages cells in a very localised part of the in does not always kill cells- sometimes it causes the occurs when a large number of cells are killed of radiation received depends on the length of expo	body	1 1 1
		(all	5 box	kes correct: 2 marks, 3 or 4 boxes correct: 1 mark);		[max 2]
	(g)	neut elec	ctron tron ; ctron con ar			[4]
		•				
						[Total: 16]

Page 5	Mark Scheme	Syllabus	Paper
	IGCSE – May/June 2013	0654	23

6 (a)

a cell formed when the nuclei of the male and female gamete fuse	zygote ;
a male gamete	sperm;
the organ in which sperms are made	testis;
the place where fertilisation occurs	oviduct;

[4]

(b) (i) 30 days ;

[1]

(ii) day 25;

[1]

(c) (i) human immunodeficiency virus;

[1]

(ii) virus passes from mother to child; across placenta; through <u>blood</u> during birth process; in breast milk;

[max 2]

[Total: 9]

7 (a) (i) good heat conductor;

malleable;

good electrical conductor;

ductile;

unreactive/does not react with water;

[max 3]

(ii) transition metals;

[1]

(b) (i) alloy;

[1]

(ii) harder/stronger;

[1]

(c) (i) electrolysis;

[1]

(ii) left electrode labelled cathode; (reject labelling of wiring or power pack) [1]

(iii) copper chloride;

gas produced is chlorine;

elements in the compound must have been copper and chlorine; [max 2]

[Total: 10]

Page 6	Mark Scheme	Syllabus	Paper
	IGCSE – May/June 2013	0654	23

\sim	/ _ \	1 - 1 - 4 - 4	_
×	121	KINATIC	٠.
8	(a)	kinetic	

heat and sound; [2]

(b) (i) liquid particles touching; gas particles not touching; random arrangement for both;

[3]

(ii) faster moving molecules;

can do more work against attractive forces/can break bonds between them; break free/separate/turn into gas/leave liquid; energy/heat (from surroundings) used for this;

[max 3]

(c) $V = I \times R$; $R = 220/3 = 73.3 (\Omega)$;

[2]

[Total: 10]

9 (a) (i)

substance	source	part of plant that absorbs it	process by which it is absorbed
carbon dioxide	air	leaf/stomata;	diffusion;
water	soil;	root/root hairs ;	

[4]

(ii) carbon dioxide + water;

glucose/sugar/starch/carbohydrate + oxygen;

[2]

(b) (i) pitchers have slippery rim (so insects fall in);

pitchers have downward-pointing spines (so insects can't crawl out);

[2]

(ii) breakdown of large molecules;

so that they can be absorbed/become soluble;

[2]

(iii) enzymes/proteases/trypsin/pepsin;

[1]

(c) (i) as a control/to make sure the only variable was the substance used;

[1]

(ii) (yes)

insects moved towards the piece of rim; use of figures from table;

[2]

[Total: 14]

	Pa	ge 7	'	Mark Scheme	Syllabus	Paper
				IGCSE – May/June 2013	0654	23
10	(a)	(i)	redu calci relea	ium carbonate uce acidity/increase pH/neutralise acids; ium carbonate reacts with/neutralises acids; ases nutrients from soil;		
			incre	assium compounds ease plant nutrient levels; assium compounds are essential for healthy plant gro	owth;	[max 3]
		(ii)		ash with dilute acid ; oon dioxide shows carbonate present ;		[2]
	(b)	(i)	15 ;			[1]
		(ii)		uric acid ; tralisation ;		[2]
	((iii)	allov	m solution gently ; w water to evaporate ; w solution to cool ;		[max 2]
						[Total: 10]
11	(a)	pea	k bet	pes up and then down again ; tween 30 and 45°C ; to by 60°C ;		[3]
	(b)	(i)	bloo refer	in environmental condition 2 od vessel is wider ; rence		
			to sv	weat ;		[2]
		(ii)	hotte	er;		[1]
		(iii)		rer/contract ; ase heat ;		[2]
						[Total: 8]