## **CAMBRIDGE INTERNATIONAL EXAMINATIONS**

**Cambridge International General Certificate of Secondary Education** 

## MARK SCHEME for the May/June 2015 series

## 0654 CO-ORDINATED SCIENCES

**0654/61** Paper 6 (Alternative to Practical), maximum raw mark 60

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

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Page 2	Mark Scheme	Syllabus	Paper
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1 (a) (i) outline concave on one side and projections on the other; [2]

2 circles shaded and labelled;

(ii) xylem; [2]

transport of water;

(b)

test solution	observation	conclusion		
Benedict's solution	orange	reducing sugar/glucose (present);		
biuret solution	blue	protein absent ;		
iodine solution	orange	starch absent ;		

[3]

(c) Any 3 from 4 [3]

(celery in dyed water and) measure distance dye moves;

minimum 3 different temperatures;

time for coloured water to appear at top of (cut) stalk/set time and measure distance moved for each T;

all other conditions/named condition kept constant;

[Total: 10]

- **2** (a) 14 and 16; [1]
  - **(b) (i)** 0.7(0) 0.8(0); [3]

0.49 and 0.64;

 $T^2$  to 2 d.p.;

Allow ecf

- (ii) 4 plots correct  $\pm$  1/2 small square ; [2] best fit straight line through origin  $\pm$  1/2 small square ;
- (iii) gradient shown clearly on graph (triangle at least 1/2 of graph); [2] 1.6;
- (iv) 39.5/gradient from (b)(iii) = 25; quoted to 2 sig figs; [2]

[Total: 10]

Page 3		Syllabus 0654	Paper 61						
3 (a) (i)	blue/pur		[1]						
(ii)	calcium h		[2]						
	calcium c								
(b) (i)	(sodium l		[3]						
	(ammonia	(ammonia) (light) blue ppt ;							
	(ammonia	(ammonia) dark blue solution (in excess);							
(ii)	(ii) CuO (not name) ;								
ado	d sodium h	et with (e.g.) sulphuric acid ; sodium hydroxide (soln)/ammonia (soln) ; e ppt (dissolves in excess) ;							
					[Total: 10]				
4 (a) (i)	(a) (i) A white blood cell; B red blood cell; C platelet; D plasma;				[4]				
(ii)	8;				[1]				
(iii)	0.008;;		[2]						
	ecf								
(b) (i)				_					
	activity	average pulse rate for 15 seconds	average heart rate (beats per minute)						
	resting	17	68						
	jogging	35	140						
(ii)	(ii) heart rate increases ;								
	increased								
	need mo								
(iii)	average	calculated / identify anomalies	average calculated/identify anomalies/confirms similar values/repeats;						

Pa	ige 4	4	Mark Scheme					Syllabus	Paper	
			Camb	ridge IG	CSE - M	lay/June	2015		0654	61
5	(a)	use of cell/ba	ttery/pow	er suppl	y and co	nnections	s ;			[3]
		connect in circ	onnect in circuit ;							
		(first two mark	first two marks can be from a diagram)							
		lamp works if lamp lights ;								
	(b)	ammeter sym	meter symbol correct and in series with lamp;							[3]
		voltmeter sym	bol corre	ct and in	parallel v	vith lamp	;			
		circuit ;								
	(c)									[3]
		(lamp)	eg A	В	С	D	E			[0]
	=	current/A								
	-	potential difference/V								
	L	table with hea	dings (all	ow p.d.) ;				<b>-</b>		
		correct units (a room for 5 land				tters, nur	nbers or	not at all ;		
	(d)	resistance = potential difference (voltage)/current;							[1]	
										[Total: 10]
6	(a)	hydrogen; lighted splint; pop (etc.);								[3]
	(b)	conical flask with delivery tube; (connected to) syringe or measuring cylinder over water;							[2]	
	(c)	(i) rate decre (then) sto								[2]
		(ii) Mg or acid or reactant(s) used up/all Mg or acid or reactant reacted					d;	[1]		
	(d)	line <b>T</b> to left of line <b>T</b> reaches								[2]

[Total: 10]