WAN, Palls

UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS

International General Certificate of Secondary Education

MARK SCHEME for the May/June 2010 question paper for the guidance of teachers

0654 CO-ORDINATED SCIENCES

0654/21

Paper 21 (Core Theory), maximum raw mark 100

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

• CIE will not enter into discussions or correspondence in connection with these mark schemes.

CIE is publishing the mark schemes for the May/June 2010 question papers for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level syllabuses and some Ordinary Level syllabuses.

www.PapaCambridge.com Page 2 Mark Scheme: Teachers' version **Syllabus** 0654 IGCSE - May/June 2010

1 (a) brain labelled;

> (b) stimuli; receptors; nerves; effectors;

(c)

	sexual reproduction	asexual reproduction
This involves gametes.	✓	
There is only one parent.		✓
The offspring are genetically identical.		✓

[3] ;;;

[Total: 8]

2 (a) (i) B – metamorphic;

C - igneous; [2]

(ii) reference to the heating (of rock A) (by rock C); [1]

(b) (i) carbon dioxide gas produced;

shows the soil contains a carbonate;

limestone is (mainly) composed of (calcium) carbonate; [max 2]

(ii) ammonia; [1]

(iii) ammonium; [1]

[Total: 7]

[3]

3 (a) (power =) work/time;

$$= 12000/60 = 200 (W);$$
 [2]

(b) (speed =) distance/time;

$$= 600/2 = 300 \, (\text{m/s});$$
 [2]

(c) (i) all symbols correct;

```
four cells displayed;
all symbols connected in series;
```

(ii) 6(V); [1]

Page 3	Mark Scheme: Teachers' version	Syllabus	V
	IGCSE – May/June 2010	0654	

(d) (density =) mass/volume; = $5/10 = 0.5 (kg/dm^3)$;

(e) two straight lines coming to a focus on the twigs/grass;

[Total: 11]

[max 2]

- **4 (a) (i)** proteins ; [1]
 - (ii) monomer(s); [1]
 - (iii) glucose; [1]
 - (b) (i) paper; building materials; furniture;

fuel; other correct;

(ii) loss of habitat;
threat to biodiversity;
threat to (new) chemical resources;
other reasonable;

[max 1]

(c) nylon, softens/melts, then hardens (on cooling);
cycle repeats on further heating;
melamine resin does not soften/it chars:

melamine resin does not soften/it chars; [3]

[Total: 9]

- 5 (a) (i) C and D; [1]
 - (ii) A and D; [1]
 - (b) (bicuspid/tricuspid/atrioventricular) valve is (pushed) shut;idea that pressure of blood causes this;[2]
 - (c) (i) haemoglobin; [1]
 - (ii) iron; [1]
 - (iii) for respiration/to combine with glucose; to release energy/to provide energy; [2]
 - (d) white blood cells fight disease; bacteria/viruses/pathogens; [2]

	Pa	ge 4	<u> </u>	Syllabus	O Tr	
	<u> </u>	.go .		Mark Scheme: Teachers' version IGCSE – May/June 2010	0654	800
	(e)		caus	lin secreted ; ses liver to absorb glucose from the blood ; changes glucose to glycogen ;		O ADAC AMBRIDGE
		` ,				Total: 13
6	(a)	(i)		nd Z/Y and Z (no mark) is acid/has pH less than 7 and Z is alkali/has pH g	reater than 7 ;	[1]
		(ii)	in m	meter immersed in one liquid as the other is added/ iixture after each addition ; rence to pH 7 ;	pH meter immersed	[2]
		(iii)	tell v acid	what ph it is / ability to tell how acidic a solution is ic;	s rather than simply	[1]
	(b)	(i)	_	nesium chloride ; ium sulfate ;		[2]
		(ii)	boili	ng/ion exchange/sodium carbonate/bath salts/wa	shing soda ;	[1]
	(c)	con	npour	contains only one type of atom ; nd contains different atoms (bonded) ; e to diagram e.g. H ₂ and O ₂ diagrams show only on	e size of circle ;	[max 2] [Total: 9]
7	(a)	foai	m, sto	r, is a poor <u>conductor/gap prevents conduction</u> ; ops <u>convection</u> of air/traps air; reflected by shiny surfaces/foil/metal;		[max 2]
	(b)	wat	er ca	n conduct electricity/danger of electrocution/electri	c shock ;	[1]
	(c)	(i)	60 W	V ;		[1]
		(ii)	incre	ease it/double it ;		[1]
		(iii)	•	ut – electrical ; out – light and heat ;;		[3]

[2]

(iv) named part of spectrum; use;

	Pa	ge 5	;	Mark Scheme: Teachers' version	Syllabus	Y.
				IGCSE – May/June 2010	0654	
	(d)	(i)	prod	rent (flows in circuit) /electricity passes through; duces (electro)magnet; attracts iron bolt;		Cambridge
		(ii)	•	no mark)minium is not magnetic/not attracted to electromagnet		[1]
		(iii)		s – no mark) an electromagnet so still attract bolt ;		[1]
		(iv)	mor	re coils/bigger voltage/bigger core;		[1]
					[To	otal: 16]
8	(a)	Gei	iger c	counter/Geiger Müller tube ;		[1]
	(b)	(i)	can	remove electrons from atoms/can form ions;		[1]
		(ii)	alph	na radiation is more ionising than gamma;		
				re likely to be absorbed by body/cells; cause more damage internally;		[max 2]
	(c)	nuc	lei sp	plit;		[1]
	(d)			ve clothing described / radiation badges to monitor g to stop radiation ;		[1] Fotal: 6]
9	(a)	(i)	root	t/root hair ;		[1]

[1]

[1]

[max2]

[max 2]

(ii) nitrogen gas is, unreactive/inert;

(iv) shortage of something in the soil;

correct ref. to function of P or K;

more proteins can be made (so more growth);

which needs to be, broken down decomposed;

NPK has ions that can be absorbed immediately;

(v) manure contains plant and animal waste e.g. proteins/urea;

to produce, ammonia/nitrates/something that can be used by plants;

detail, e.g. more cells/more cytoplasm;

(iii) to make protein/amino acids;

	Pa	ige 6	<u> </u>		M	ark Sc	heme	: Teac	hers'	versio	n	Syllabus	7.0 T
		9						May/J				0654	Star
	(b)	• • •			; ioxide	;							MA. Parla Cambridge
		(iii)	palis	sade/	meso	phyll	· ;						[1]
		(iv)	by d tran	ugh s liffusion spirate porati	tion ;	ta ;							[max 2]
													[Total: 13]
10	(a)		,	Υ;			/			x ;			
													[2]
	(b)		heat	as a d ted w ned)	cataly rith ca more	st ; irbon / reactiv	carboi ve me	n mono tal ;	oxide/	-	educer	ploured compound r which works e.g	ds / can [1]
	(c)	(i)				e/CO ₂ e/H ₂ O							[2]

(ii) produces hotter flame/reaches a higher temperature;

reasonable reference to air behaving as 'dilute' oxygen; reference to higher temperatures needed to melt metals;

[max 1]

[Total: 8]