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## UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS

**International General Certificate of Secondary Education** 

## MARK SCHEME for the May/June 2010 question paper for the guidance of teachers

## 0654 CO-ORDINATED SCIENCES

0654/31

Paper 31 (Extended Theory), maximum raw mark 100

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

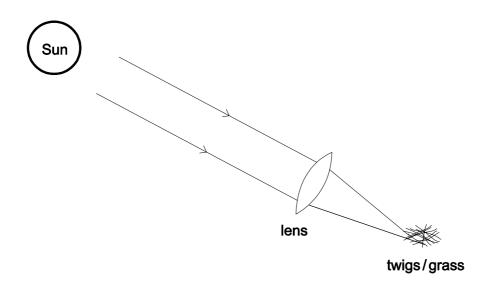
• CIE will not enter into discussions or correspondence in connection with these mark schemes.

CIE is publishing the mark schemes for the May/June 2010 question papers for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level syllabuses and some Ordinary Level syllabuses.

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Page 2	Mark Scheme: Teachers' version	Syllabus
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(a)	(i)	C and D;	Cambridge
	(ii)	A and D;	TO
	(iii)	opens and closes; when atrium contracts valve is pushed open; when ventricle contracts valve is pushed shut;	[max 2]
(b)	idea	re oxygen (in right side of heart in fetus); a that it is a mix of oxygenated blood (from placenta) and deoxygenated m body tissues);	[2]
(c)	(i)	haemoglobin ;	[1]
	(ii)	protein;	[1]
	(iii)	iron;	[1]
	(iv)	small particles/not made of large molecules; so can be absorbed as they are;	[2]
	(v)	for respiration/to combine with glucose; to release energy/to provide energy;	[2]
			[Total: 13]
(a)	(i)	causes, skin cancer/eye damage/burns/mutation in skin/damage to DNA in skin;	[1]
	(ii)	protective clothing/sun block;	[1]
(b)		eed =) distance/time ; 00 m/s ;	[2]
(c)		mentum =) mass × velocity ; 00 000 × 60 = 24 000 000 kg m/s ;	[2]
(d)	(i)	all symbols correct; all symbols connected in series;	[2]
	(ii)		- 43
		6V;	[1]

Page 3	Mark Scheme: Teachers' version	Syllabus
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two stra	ight parallel rays drawn entering the lens; ight rays brought to a focus at the twigs/grass; correctly shown;	andridge.c
Sur		Oh Oh



[Total: 15]

3 (a) ionising;

removes electrons;

damages DNA/mutation;

effect (e.g. cancer/burns/radiation sickness);

[max 3]

(b) (i) <u>nuclei</u> split/<u>nuclear</u> fission;

[1]

(ii) nuclear/radioactive/toxic waste; problems of disposal/storage;

security of fissionable/radioactive material;

use in terrorism;

accident/malfunction;

effect of radioactive materials on environment/humans;

[max 2]

[Total: 6]

(a) (i) reaction is exothermic/heat was given off;

[1]

(ii) temperature falls (after 25 cm<sup>3</sup> of acid added);

so no further (exothermic) reaction/all alkali used up;

[2]

		J	IGCSE – May/June 2010 0654	20
	(b)	(i)	moles of A ((25.0/1000) × 0.2 =) 0.005;	a Cambridge
			moles KOH $((20.0/1000) \times 0.5 =) 0.01$ ; (allow 1 mark if the same error in converting to dm³ is made in each calculation, e.g. if left in cm³ answers are 5 and 10)	[2]
		(ii)	(0.5) (no mark) [e.c.f. from (i) provided answer is <u>half</u> the KOH moles] because the number of moles of acid must be half the number of moles of KOH / owtte / or relevant working;	[1]
		(iii)	$H^+ + OH^- \rightarrow H_2O$ (all correct for 2 marks, two of the three for 1 mark) ;;	[2]
	(c)	(i)	electrolysis;	[1]
		(ii)	plate, has a negative charge/is negative, and potassium ions, are positively charged/are positive; opposite charges attract/potassium ions move towards the plate; potassium ions gain electrons from the plate;	
			potassium ions, discharged/gain one electron/become atoms;	[max 3]
			רו	Total: 12]
5	(a)	foai	m/air, is a poor <u>conductor</u> ; m, stops <u>convection</u> of air/traps air; <u>iation</u> reflected by, shiny surfaces/foil/metal;	[3]
	(b)	(i)	<b>B</b> (no mark) turns ratio 2:1;	[1]
		(ii)	water can conduct electricity; danger of electrocution;	[2]
	(c)	(i)	current (flows in circuit); produces (electro)magnet; (magnet) attracts iron bolt;	[3]
		(ii)	(no – no mark) aluminium is not magnetic/not attracted to electromagnet;	[1]
		(iii)	(yes – no mark) still an electromagnet (so still attracts bolt) ;	[1]
		(iv)	more coils/bigger voltage/bigger core;	[1]

Mark Scheme: Teachers' version

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Syllabus

[Total: 12]

Page 5		<u> </u>	Mark Scheme: Teachers' version Syllabus		Q V	
			IGCSE – May/June 2010 0654	0654	ADS.	
6	(a)	(i)	amn	nonium/NH₄ <sup>+</sup> ;	·	Cally .
		(ii)	nitro prote deta	rtage of something in the soil; ligen/nitrate, needed for making, protein/amino acid leins for growth; lil, e.g. more cells/more cytoplasm; lect ref. to function of P or K;	ls;	[max 3]
		(iii)		at – little/no, difference ;		
			•	itoes – greater, with manure + bacteria/in plot <b>B</b> ; to tonnes (per hectare per year) (greater);		[3]
		(iv)	which to pr	nure contains plant and animal waste e.g. proteins/uch needs to be, broken down/decomposed (by bacteroduce, ammonia/nitrates/something that can be use rence to nitrification/nitrifying bacteria;	eria) ;	[max 2]
	(b)	plaı fed whi	nts/a on by ch re	es growth of, algae/plants ; lgae, die ; y bacteria/decomposers ; spire (aerobically) ;		[mov 2]
		bac	teria	use oxygen ;		[max 3]
						[Total: 12]
7	(a)	(i)	gluc	ose;		[1]
		(ii)		ein; proteins contain, S/sulfur ; proteins contain, N/nitrogen ;		[3]
	(b)	(i)	there the limit molt	ecules have only weak forces between them; ecules/particles, can move past one another easily efore (solid) nylon, melts / becomes a liquid, whe hot container; een nylon can be pumped (through small holes); een threads solidify when cooled; ng forces between molecules when solid;	-	s [max 3]
		(ii)	does	sn't melt (on contact with hot containers); ecules cannot move past one another;		
				ause strong bonds hold polymer chains/crosslinks;		[max 2]

[Total: 9]

[clear diagram could score crosslink mark]

	Page 6	Mark Scheme: Teachers' version	Syllabus
	-	IGCSE – May/June 2010	0654
3	(a) A to reting B to option C to iris	ic nerve;	Cambridge C
		uscles, contract/get shorter ; tension on) (suspensory) ligaments ;	OH)

(b) ciliary muscles, contract/get shorter; loosen (tension on) (suspensory) ligaments; lens more rounded/fatter; more refraction/shorter focal length; light (rays) brought to a focus on the retina;

[max 3]

(c) cystic fibrosis/sickle cell anaemia/thalassaemia/other; statement as to whether allele is dominant or recessive; (above examples are all recessive. Huntington's is dominant)

## if recessive

both parents must have allele for offspring to inherit disease/are heterozygous; parental genotypes and offspring genotypes shown / 1 in 4 chance of offspring having disease;

or

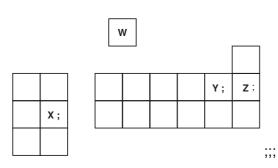
if dominant

only one parent needs to have allele for offspring to inherit disease; parental genotypes and offspring genotypes shown / 1 in 2 chance of offspring having disease;

[max 3]

[Total: 9]

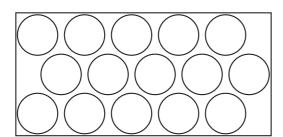
9 (a)



[3]

(b) (i) atoms all same size arranged in regular lattice; e.g.

[1]



(ii) reference to delocalised electrons; movement of charge/electrons;

[2]

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(c) (i) oxidation/reaction with oxygen (from air)/formation of metal oxide; reference to the, hot/molten, metal;

(ii) three shells with 18 electrons; arranged 2,8,8;

2]

(iii) outer shell is complete; does not need to, lose/gain electrons, (by reaction)/owtte;

[2]

[Total: 12]