UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS International General Certificate of Secondary Education

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for the guidance of teachers

0654 CO-ORDINATED SCIENCES

0654/21

Paper 2 (Core Theory), maximum raw mark 100

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

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<u> </u>	age 2	Mark Scheme: Teachers' version Syllabus	2
		IGCSE – May/June 2011 0654	Day
(a)	one	el/an alloy is a mixture (of metals and other elements)/contains more that e element ; ld steel) contains carbon (mixed with iron) ;	Dana cannonios an Imprios
(b)) (i)	in B , air/oxygen and water are present (together)/air and water needed for rusting ; no water in A ;	or
		no air / oxygen in C ;	[3]
	(ii)	oxidation ;	[1]
	(iii)	not enough air / oxygen present / only water present ;	[1]
(c)	i) (i)	W and Y ; contain only hydrogen and carbon ;	[2]
	(ii)	does not mix with water / air / oxygen ; sticks to chain / steel ;	[max 1]
(d)		ymer molecule much larger / longer / heavier ; a that polymer is made from simple molecules / monomers linked into chain ;	[max 1] [Total: 10]
(a)) (i)	number of waves per second ;	[1]
	(ii)	(distance =) speed × time ; = 300 000 000 × 0.000 027 = 8100 (m) ; so distance = 4050 (m) ;	[3]
(b)		$E =) \frac{1}{2} mv^2$; $E \times 140000 \times 100 \times 100 = 7 \times 10^8 (J)$;	[2]
(c)) (i)	C = weight, D = drag / friction / air resistance ;	[1]
	(ii)	constant speed/no acceleration (means balanced forces);	[1]
(d)		celeration =) change in velocity/time ; 5/40 or 2.125 (m/s²) ;	[2]

	3 Mark Scheme: Teachers' version Syllab	bus S. r
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	eceptor) nose / cells in nose ; fector) salivary glands ;	bus 4 and combine
(b) (i)	catalyst ; protein ; speeds up/controls/catalyses, metabolic reactions ;	[max 2]
(ii)	to break down/digest starch to ; sugar/maltose ; that can be absorbed/that can move from gut into the blood ;	[max 2]
		[
(c) (i)	grinding / crushing ; increase surface area of food ; idea of easier access for enzymes ;	[max 2]
(ii)	-	
	cells ;	[max 3]
(iii)	contains calcium ; needed to form enamel ;	[2]
		[Total: 13]
(a) (i)	electrons ;	[1]
(ii)	negative ;	[1]
(iii)	length ; temperature ; cross sectional area/width/diameter ; material/resistivity/conductivity ;	[max 2]
(b) (i)	red, green and blue ;;	[2]
(ii)	other colours produced by a combination of these ;	[1]
(c) (i)	heat/thermal;	[1]
(0) (1)		[1]
(c) (i) (ii)	increase temperature / produce convection current ;	L'J
		[2]

 Pa	ge 4	Mark Scheme: Teachers' version Syllabus	Par I
 		IGCSE – May/June 2011 0654	230
(a)		amics ; prine ; ss ; er ;	Papa Camping
(b)	(i) (ii)	compound has formula / fixed proportions of elements ; compound has different elements bonded together ; compound has different properties from constituents ; (significant) energy change when compound formed ; (or corresponding statements for mixture) <u>fractional distillation</u> ;	[max 2] [1]
(c)	incr	ease temperature ; ease pressure ; catalyst ;	[max 2]
(d)	acio neu	l ; tralisation ;	[2]
			[Total: 11]
(a)	(i)	23 ; chromosomes ;	[2]
	(ii)	label to cell membrane ; label to cytoplasm ;	[2]
((iii)	pointed head, reduces friction/streamlined ; tail for swimming ;	[2]
(b)	test	is/testicle ;	[1]
(c)	(i)	oxygen use by one sperm/single sperm quantities too small to measure ;	[1]
	(ii)	respiration ; oxygen combined with sugar to release energy ;	
		more energy used when swimming ;	[2 max]

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(a) diag	ram showing second swi	itch in parallel with first ;		www.papac	ambrid
(b)	switch X	switch Y	lamp off or on		30
	up	up	on		
	up	down	off		
	down	цр	off		
	down	down	on		
(c) (i)	heated water rises/cold by convection ;			,,	[2]
				,,	[2] [max 2] [1]
(ii) (d) (larg (stro	by convection ; hot water less dense / co	ld water more dense ; ng electromagnet ;		,,	[max 2]
(ii) (d) (larg (stro cont	by convection ; hot water less dense / co 5000 (J) ; ge current produces) stro ong enough to) attract iro	ld water more dense ; ng electromagnet ;		"	[max 2] [1]
(ii) (d) (larg (stro cont (e) (i) (ii)	by convection ; hot water less dense / co 5000 (J) ; ge current produces) stro ong enough to) attract iro tacts break ; coal / oil / gas / peat ;	ld water more dense ; ng electromagnet ;	varming/no use of		[max 2] [1] [3]
(ii) (d) (larg (stro cont (e) (i) (ii) (iii)	by convection ; hot water less dense / co 5000 (J) ; ge current produces) stro ong enough to) attract iro tacts break ; coal / oil / gas / peat ; no CO ₂ emissions / no fuels / renewable ;	old water more dense ; ng electromagnet ; n (on pivot) ; o addition to global v ines noisy/can't work	-	fossil	[max 2] [1] [3] [1]

Pa	ge 6	Mark Scheme: Teachers' version Syllabus	L.
	<u>.</u>	IGCSE – May/June 2011 0654	2
(a)	cark stor	Mark Scheme: Teachers' version Syllabus IGCSE – May/June 2011 0654 biration ; 000 dioxide ; mata ; tosynthesis ;	Sambids
(b)	use	ates absorbed by plant roots ; d for making proteins ; æins used for making new cells ;	[max 2]
(c)	(i)	to kill / destroy, insects ; which eat / damage, crop / grass for grazing ; increase yields ;	[max 2]
	(ii)	kill dung beetles ; so dung not buried / nitrates in dung do not get into soil ;	[2]
		[Total: 10]
(a)	(i)	13 ;	[1]
	(ii)	potassium feldspar ; lilac shows potassium ;	[2]
((iii)	calcium / potassium ;	[1]
(b)	(i)	no wind for sandblasting ; no water for freeze / thaw ; no water for chemical weathering ; no plants / animals for biological weathering ;	[max 2]
	(ii)	salts / minerals are released into the soil ; which plants need for healthy growth / maintenance ;	[2]
(c)	(i)	(thermal) decomposition ; (heating) causes a substance to break down into simpler ones/a large substance is broken down into smaller ones/calcium oxide (and carbon dioxide) is (are) simpler substances than calcium carbonate ;	[2]
	(ii)	(calcium oxide has lower mass) mass due to carbon dioxide has been lost/part of the calcium carbonate has been lost/calcium oxide is only a part of calcium carbonate ;	[1]
((iii)	green to blue / purple ; reaction produces an alkali / alkaline solution / calcium hydroxide ;	[2]