



UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS  
International General Certificate of Secondary Education

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**CO-ORDINATED SCIENCES**

**0654/11**

Paper 1 Multiple Choice

**May/June 2012**

**45 minutes**

Additional Materials: Multiple Choice Answer Sheet  
Soft clean eraser  
Soft pencil (type B or HB is recommended)

\* 4 3 8 6 2 5 7 4 2 4 \*

**READ THESE INSTRUCTIONS FIRST**

Write in soft pencil.  
Do not use staples, paper clips, highlighters, glue or correction fluid.  
Write your name, Centre number and candidate number on the Answer Sheet in the spaces provided unless this has been done for you.

There are **forty** questions on this paper. Answer **all** questions. For each question there are four possible answers **A, B, C** and **D**.  
Choose the **one** you consider correct and record your choice in **soft pencil** on the separate Answer Sheet.

**Read the instructions on the Answer Sheet very carefully.**

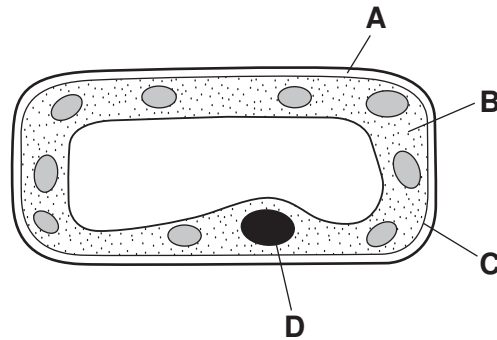
Each correct answer will score one mark. A mark will not be deducted for a wrong answer.  
Any rough working should be done in this booklet.  
A copy of the Periodic Table is printed on page 20.

This document consists of **17** printed pages and **3** blank pages.



1 The diagram shows a section through a cell from a leaf.

Which part is the cell membrane?



2 What happens in photosynthesis?

- A Carbon dioxide is made.
- B Oxygen is used.
- C Starch is absorbed.
- D Water is used.

3 Which word equation represents aerobic respiration?

- A carbon dioxide + oxygen → glucose + water
- B carbon dioxide + water → glucose + oxygen
- C glucose + oxygen → carbon dioxide + water
- D glucose + oxygen → lactic acid

4 Some cancer treatments cause a reduction in the number of a person's white blood cells.

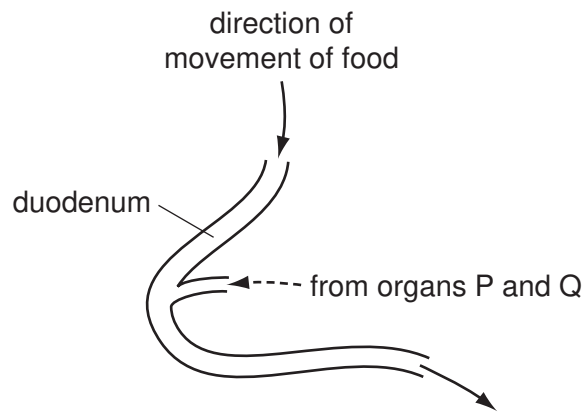
Why might this be a problem?

- A Blood takes longer to clot.
- B Infections are more likely to cause illness.
- C Insufficient oxygen reaches the brain.
- D Less carbon dioxide is carried to the lungs.

5 Why is calcium needed in the diet?

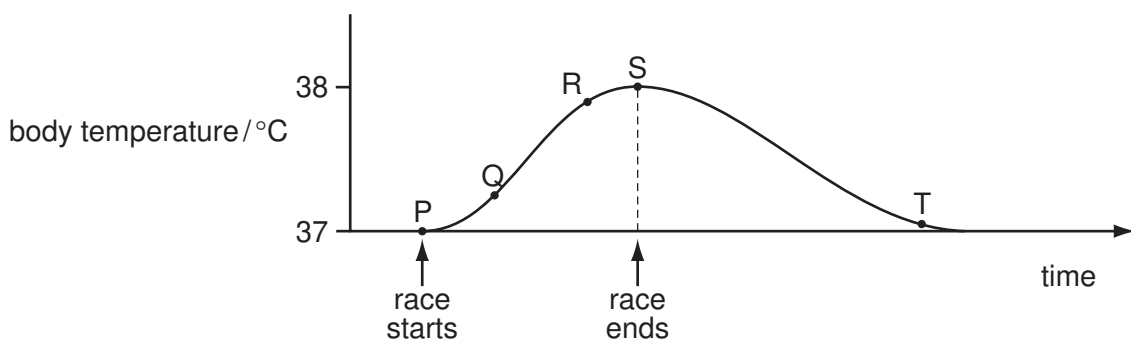
- A to make carbohydrates
- B to make teeth
- C to make enzymes
- D to make muscles hard

- 6 The diagram shows part of the alimentary canal.



Which organs are represented by P and Q?

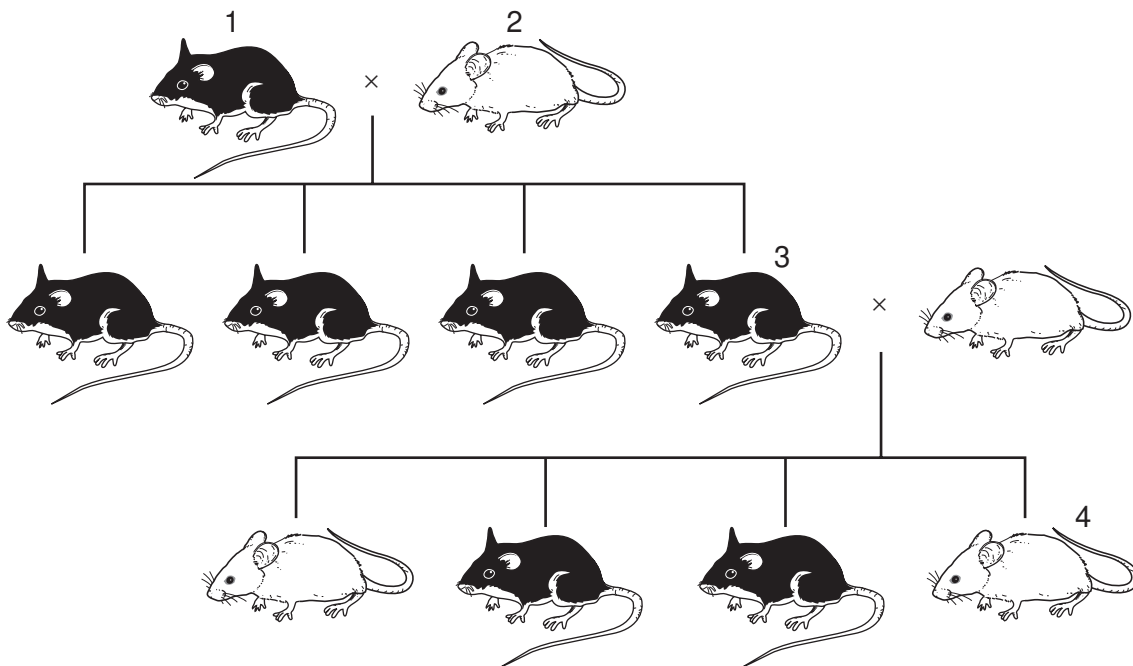
- A kidneys and pancreas
  - B liver and pancreas
  - C liver and stomach
  - D pancreas and stomach
- 7 A person touches a hot object which triggers a reflex action.  
In which order does the signal travel in the reflex arc?
- A relay neurone → spinal cord → sensory neurone
  - B sensory neurone → spinal cord → motor neurone
  - C spinal cord → sensory neurone → stimulus
  - D stimulus → motor neurone → spinal cord
- 8 The graph shows body temperature before, during and after running a race on a hot day.



Which change in body temperature occurs as a result of homeostasis?

- A P to Q
- B Q to R
- C R to S
- D S to T

- 9 Which structure contracts to expel the baby during birth?
- A cervix  
B oviduct  
C uterus wall  
D vagina
- 10 In a flowering plant, which structure contains the female gamete?
- A anther  
B ovule  
C pollen grain  
D stigma
- 11 The diagram shows the results of a breeding experiment using black and white mice.



Which statement is correct?

- A Mouse 1 has a dominant allele for fur colour.  
B Mouse 2 is heterozygous for fur colour.  
C Mouse 3 is homozygous for fur colour.  
D Mouse 4 is heterozygous for fur colour.

12 The diagram shows a food chain.

Which organisms pass the greatest amount of energy along the food chain?

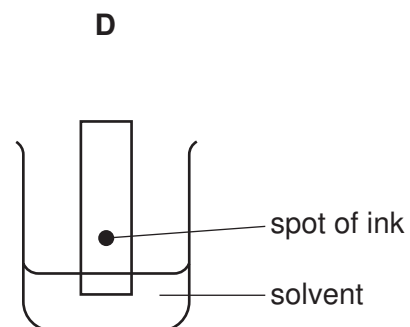
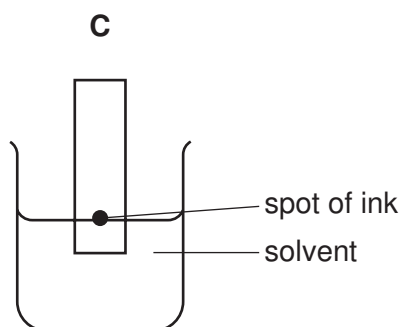
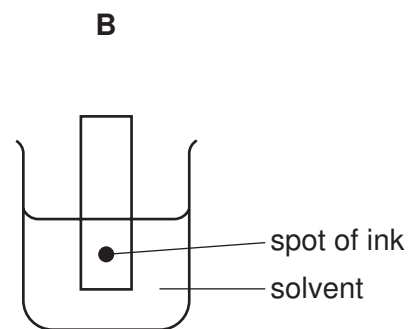
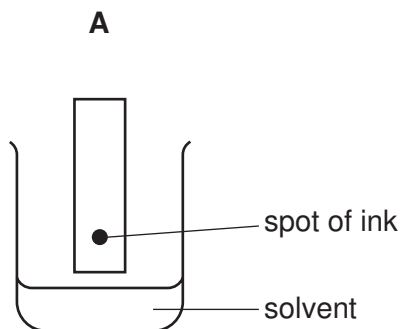


13 What can lead to global warming?

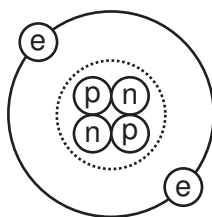
	deforestation	burning of fossil fuels
<b>A</b>	✓	✓
<b>B</b>	✓	✗
<b>C</b>	✗	✓
<b>D</b>	✗	✗

14 The colours in an ink can be separated by chromatography.

Which diagram shows the correct way to set up the apparatus?



15 The diagram shows a helium atom.



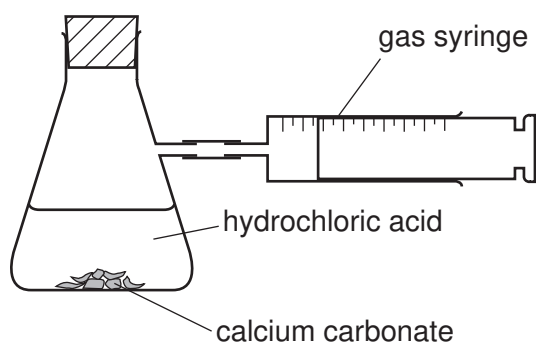
Which particles in the helium atom have approximately the same mass?

- A electron and proton only
- B electron and neutron only
- C proton and neutron only
- D electron, proton and neutron

16 How many atoms of metals and of non-metals are shown in the formula  $\text{Na}_2\text{SO}_4$ ?

	atoms of metals	atoms of non-metals
A	1	1
B	1	2
C	2	4
D	2	5

- 17 The apparatus shown is used to investigate the speed of reaction between hydrochloric acid and calcium carbonate.



The time to collect  $50\text{ cm}^3$  of gas is measured.

Using concentrated acid and lumps of calcium carbonate, the time is 150 s.

In a second experiment, the time is 90 s.

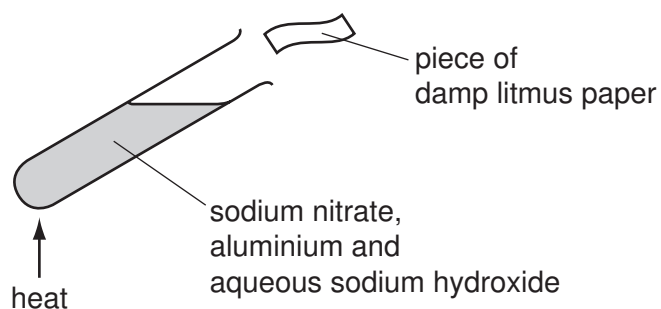
Which change was made in the second experiment?

- A larger lumps of calcium carbonate
  - B less concentrated acid
  - C lower temperature
  - D powdered calcium carbonate
- 18 Hydrogen and oxygen react explosively to form water.

Which terms describe this reaction?

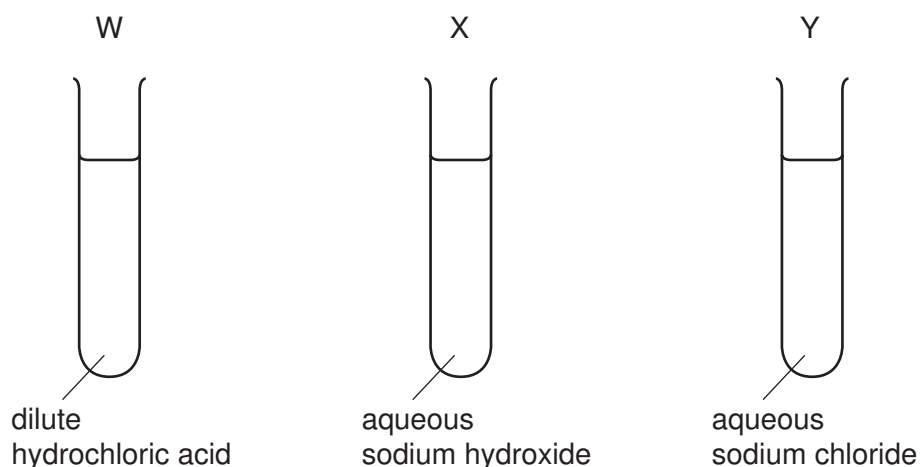
	combustion	oxidation
<b>A</b>	✓	✓
<b>B</b>	✓	x
<b>C</b>	x	✓
<b>D</b>	x	x

- 19 The diagram shows litmus paper testing the gas that is given off from the contents of the test tube.



The damp litmus paper

- A** turns blue.  
**B** turns colourless.  
**C** turns red.  
**D** turns red then colourless.
- 20 Universal Indicator solution is added to test-tubes W, X and Y.



What are the colours of the Universal Indicator?

	in W	in X	in Y
<b>A</b>	green	red	purple
<b>B</b>	purple	green	red
<b>C</b>	red	green	purple
<b>D</b>	red	purple	green



21 The table shows physical properties of some substances.

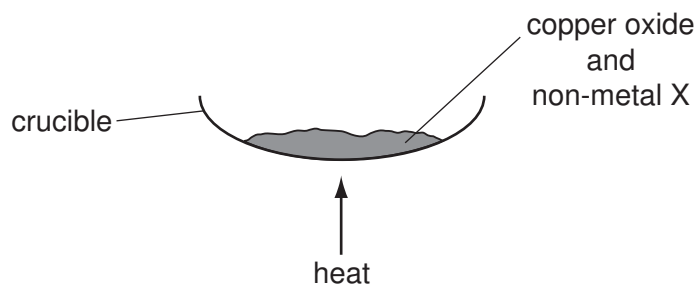
Which substance is a metal?

	malleability	density	electrical conductivity
<b>A</b>	brittle	high density	high
<b>B</b>	brittle	low density	low
<b>C</b>	malleable	high density	high
<b>D</b>	malleable	low density	low

22 Which statement about lithium, sodium and potassium is **not** correct?

- A** They are in the same group of the Periodic Table.
- B** They are in the same period of the Periodic Table.
- C** They float on water.
- D** They react with water to give a flammable gas.

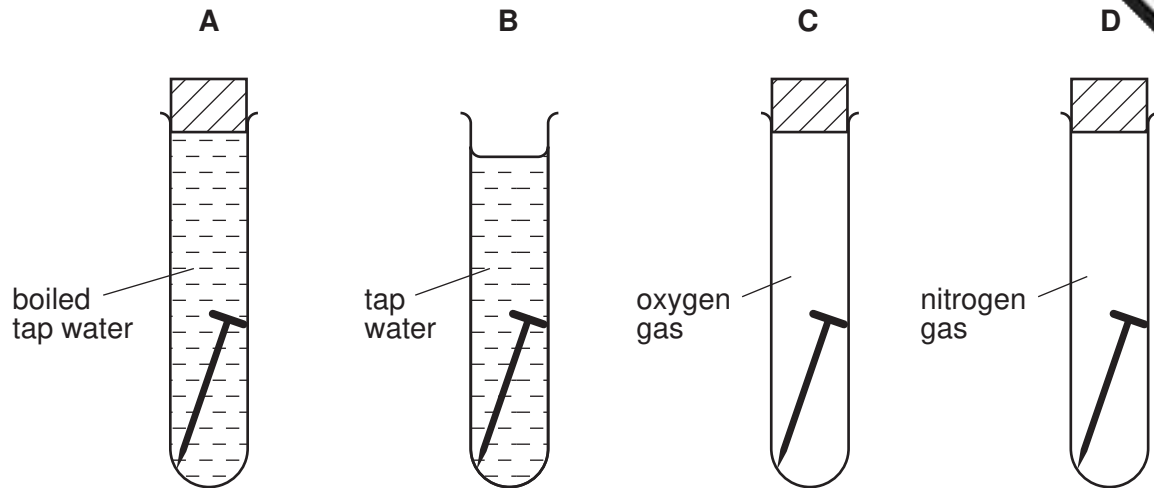
23 Copper is obtained from copper oxide by heating with non-metal X.



Which shows the identity of non-metal X and the type of reaction non-metal X undergoes?

	identity of X	type of reaction
<b>A</b>	carbon	oxidation
<b>B</b>	carbon	reduction
<b>C</b>	oxygen	oxidation
<b>D</b>	oxygen	reduction

24 In which tube does the iron nail rust in the shortest time?

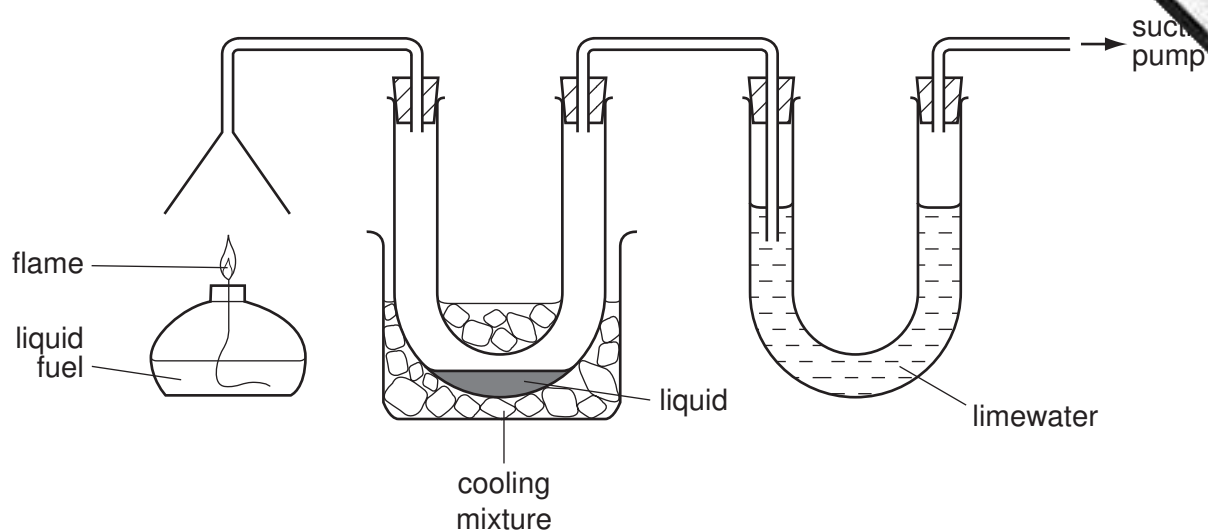


25 Fertilisers are used to supply the essential elements needed for plant growth.

Which compound supplies two of these essential elements?

- A  $\text{Ca}(\text{H}_2\text{PO}_4)_2$
- B  $\text{Ca}(\text{NO}_3)_2$
- C  $\text{KNO}_3$
- D  $(\text{NH}_4)_2\text{SO}_4$

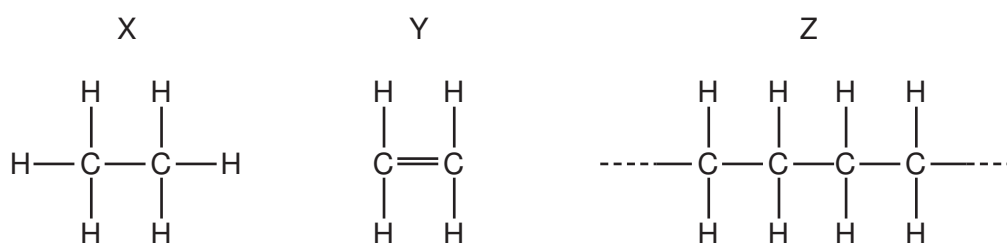
26 The burning of a fuel is investigated using the apparatus shown.



Which substances is the apparatus testing for?

- A carbon monoxide and carbon dioxide
- B carbon monoxide and water
- C carbon dioxide and water
- D carbon dioxide and sulfur dioxide

27 The diagram shows three molecules.



Which molecule is a monomer and which is a polymer?

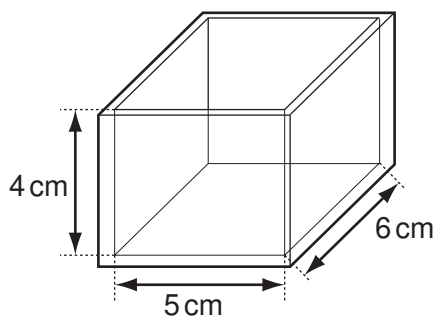
	monomer	polymer
<b>A</b>	X	Z
<b>B</b>	Y	Z
<b>C</b>	Y	X
<b>D</b>	Z	Y

- 28 A motorist starts out on a 210 km journey at 8 am. At 10 am he stops for a 30 minutes covering 180 km. The motorist completes the journey at 11 am.

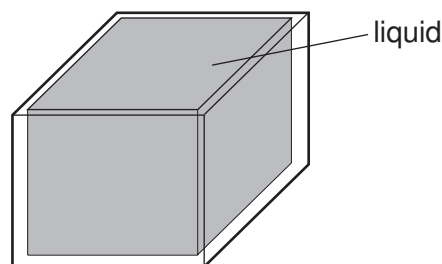
What is his average speed in covering the 210 km?

- A 60 km/h      B 70 km/h      C 84 km/h      D 90 km/h

- 29 The diagrams show a glass tank with inside measurements of 5 cm × 6 cm × 4 cm.



mass = 40 g



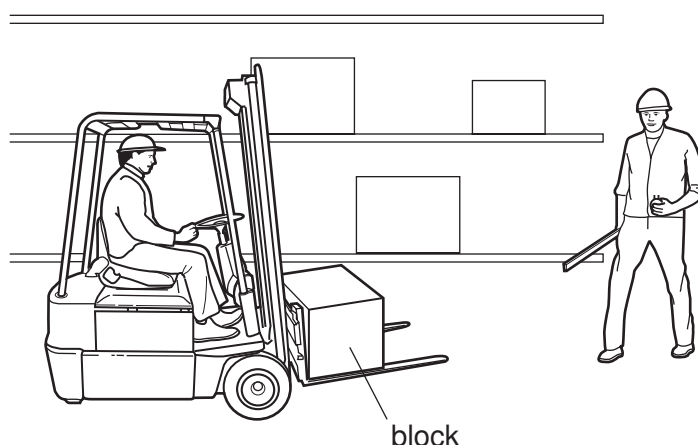
total mass = 220 g

The tank has a mass of 40 g when empty. When the tank is filled with a liquid, the tank and liquid have a total mass of 220 g.

What is the density of the liquid?

- A  $\frac{220}{(5 \times 6 \times 4)} \text{ g/cm}^3$   
 B  $\frac{(220 - 40)}{(5 \times 6 \times 4)} \text{ g/cm}^3$   
 C  $\frac{(5 \times 6 \times 4)}{220} \text{ g/cm}^3$   
 D  $\frac{(5 \times 6 \times 4)}{(220 - 40)} \text{ g/cm}^3$

- 30 A workman lifts a cubic block from ground level to a high shelf using a fork lift truck. A second workman has a metre rule and a stopwatch.



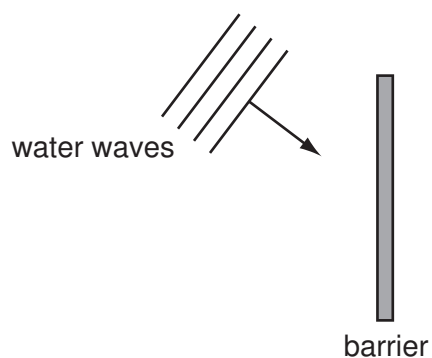
Which quantity will the second workman be able to determine, using **only** the metre rule and the stopwatch?

- A the average speed of the block as it moves up
  - B the density of the material of the block
  - C the pressure exerted by the block on the shelf
  - D the work done on the block when it is lifted
- 31 On a warm day, a driver checks the air pressure in a car tyre. Overnight the temperature drops and the air pressure in the tyre falls. There are no air leaks in the tyre.

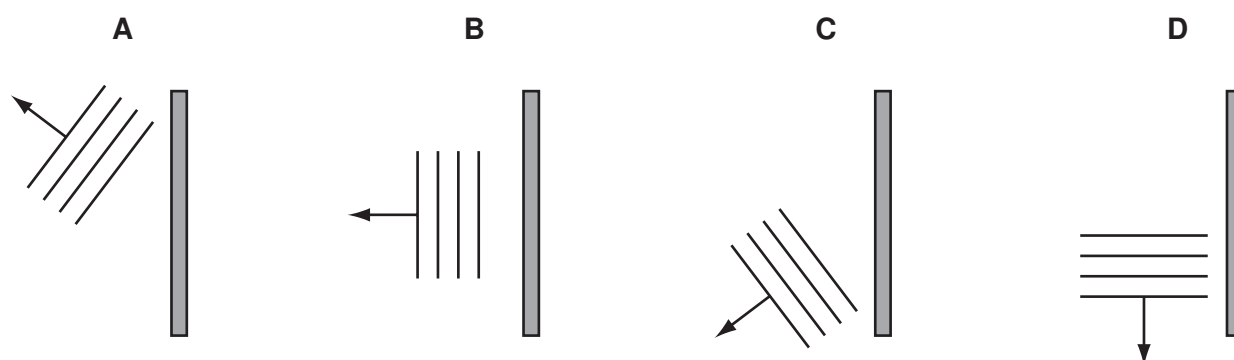
Why does the pressure fall?

- A The air molecules in the tyre move more slowly.
  - B The air molecules in the tyre stop moving.
  - C The volume of the air in the tyre decreases.
  - D The volume of the air in the tyre increases.
- 32 How is heat transferred in a vacuum?
- A by conduction and convection
  - B by convection and radiation
  - C by convection only
  - D by radiation only

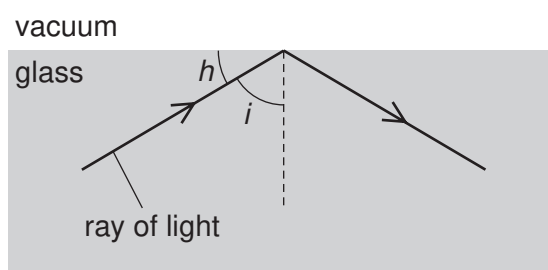
- 33 The diagram shows water waves travelling towards a barrier.



Which diagram shows the direction of the waves after being reflected by the barrier?



- 34 A glass block is surrounded by a vacuum. A ray of light strikes the inside of the glass block, and is totally reflected back into the block.



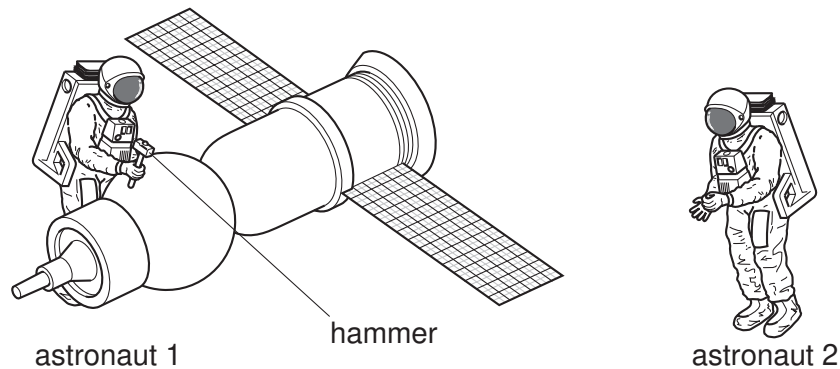
Why does this happen?

- A Angle  $h$  is greater than the critical angle.
- B Angle  $i$  is greater than the critical angle.
- C Light cannot travel through a vacuum.
- D The ray is travelling along the normal.

- 35 The Sun emits infra-red radiation, ultraviolet radiation and visible light.

Which statement about the time it takes these radiations to reach Earth's atmosphere is correct?

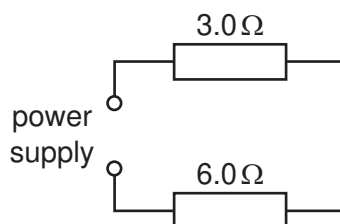
- A Infra-red radiation arrives first.
  - B Ultraviolet radiation arrives first.
  - C Visible light arrives first.
  - D They all arrive at the same time.
- 36 Astronaut 1 uses a hammer to mend a satellite in space. Astronaut 2 is nearby. There is no air in space.



Compared with the sound heard if they were working on Earth, what does astronaut 2 hear?

- A a louder sound
  - B a quieter sound
  - C a sound of the same loudness
  - D no sound at all
- 37 The instructions for a household lamp state that the plug should be fitted with a 3 A fuse.
- What could happen if, by mistake, a 13 A fuse is fitted?
- A The fuse might melt too easily.
  - B The lamp might explode if a fault develops.
  - C The wires connecting the lamp to the plug might overheat if a fault developed.
  - D Too much voltage might be supplied to the lamp.

- 38 A  $3.0\ \Omega$  resistor and a  $6.0\ \Omega$  resistor are connected to a power supply as shown.



What is the total resistance of the circuit?

- A**  $2.0\ \Omega$       **B**  $3.0\ \Omega$       **C**  $9.0\ \Omega$       **D**  $18\ \Omega$
- 39 In the lighting circuit in a house, how are lamps usually connected, and what is one reason for this?

	usual connection	reason
<b>A</b>	parallel	to allow every lamp to have the full supply voltage
<b>B</b>	parallel	to share out the voltage equally between the lamps
<b>C</b>	series	to allow every lamp to have the full supply voltage
<b>D</b>	series	to share out the voltage equally between the lamps

- 40 What are carbon-12 and carbon-14?
- A** atoms of different elements with different nuclear masses
- B** atoms of different elements with the same nuclear mass
- C** atoms of the same element with different nuclear masses
- D** atoms of the same element with the same nuclear mass









## DATA SHEET

### The Periodic Table of the Elements

Group																			
I	II											III	IV	V	VI	VII	0		
												<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     1 <b>H</b> Hydrogen 1                 </div>						<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     4 <b>He</b> Helium 2                 </div>	
<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     7 <b>Li</b> Lithium 3                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     9 <b>Be</b> Beryllium 4                 </div>											<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     11 <b>B</b> Boron 5                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     12 <b>C</b> Carbon 6                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     14 <b>N</b> Nitrogen 7                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     16 <b>O</b> Oxygen 8                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     19 <b>F</b> Fluorine 9                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     20 <b>Ne</b> Neon 10                 </div>		
<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     23 <b>Na</b> Sodium 11                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     24 <b>Mg</b> Magnesium 12                 </div>											<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     27 <b>Al</b> Aluminium 13                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     28 <b>Si</b> Silicon 14                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     31 <b>P</b> Phosphorus 15                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     32 <b>S</b> Sulfur 16                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     35.5 <b>Cl</b> Chlorine 17                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     40 <b>Ar</b> Argon 18                 </div>		
<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     39 <b>K</b> Potassium 19                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     40 <b>Ca</b> Calcium 20                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     45 <b>Sc</b> Scandium 21                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     48 <b>Ti</b> Titanium 22                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     51 <b>V</b> Vanadium 23                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     52 <b>Cr</b> Chromium 24                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     55 <b>Mn</b> Manganese 25                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     56 <b>Fe</b> Iron 26                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     59 <b>Co</b> Cobalt 27                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     59 <b>Ni</b> Nickel 28                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     64 <b>Cu</b> Copper 29                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     65 <b>Zn</b> Zinc 30                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     70 <b>Ga</b> Gallium 31                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     73 <b>Ge</b> Germanium 32                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     75 <b>As</b> Arsenic 33                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     79 <b>Se</b> Selenium 34                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     80 <b>Br</b> Bromine 35                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     84 <b>Kr</b> Krypton 36                 </div>		
<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     85 <b>Rb</b> Rubidium 37                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     88 <b>Sr</b> Strontium 38                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     89 <b>Y</b> Yttrium 39                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     91 <b>Zr</b> Zirconium 40                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     93 <b>Nb</b> Niobium 41                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     96 <b>Mo</b> Molybdenum 42                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     96 <b>Tc</b> Technetium 43                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     101 <b>Ru</b> Ruthenium 44                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     103 <b>Rh</b> Rhodium 45                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     106 <b>Pd</b> Palladium 46                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     108 <b>Ag</b> Silver 47                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     112 <b>Cd</b> Cadmium 48                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     115 <b>In</b> Indium 49                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     119 <b>Sn</b> Tin 50                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     122 <b>Sb</b> Antimony 51                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     128 <b>Te</b> Tellurium 52                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     127 <b>I</b> Iodine 53                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     131 <b>Xe</b> Xenon 54                 </div>		
<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     133 <b>Cs</b> Caesium 55                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     137 <b>Ba</b> Barium 56                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     139 <b>La</b> Lanthanum 57 *                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     178 <b>Hf</b> Hafnium 72                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     181 <b>Ta</b> Tantalum 73                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     184 <b>W</b> Tungsten 74                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     186 <b>Re</b> Rhenium 75                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     190 <b>Os</b> Osmium 76                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     192 <b>Ir</b> Iridium 77                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     195 <b>Pt</b> Platinum 78                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     197 <b>Au</b> Gold 79                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     201 <b>Hg</b> Mercury 80                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     204 <b>Tl</b> Thallium 81                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     207 <b>Pb</b> Lead 82                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     209 <b>Bi</b> Bismuth 83                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     210 <b>Po</b> Polonium 84                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     210 <b>At</b> Astatine 85                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     210 <b>Rn</b> Radon 86                 </div>		
<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     87 <b>Fr</b> Francium 87                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     226 <b>Ra</b> Radium 88                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     227 <b>Ac</b> Actinium 89 †                 </div>																	

\*58-71 Lanthanoid series

†90-103 Actinoid series

a	a = relative atomic mass
<b>X</b>	X = atomic symbol
b	b = proton (atomic) number

<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     140 <b>Ce</b> Cerium 58                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     141 <b>Pr</b> Praseodymium 59                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     144 <b>Nd</b> Neodymium 60                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     147 <b>Pm</b> Promethium 61                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     150 <b>Sm</b> Samarium 62                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     152 <b>Eu</b> Europium 63                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     157 <b>Gd</b> Gadolinium 64                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     159 <b>Tb</b> Terbium 65                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     162 <b>Dy</b> Dysprosium 66                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     165 <b>Ho</b> Holmium 67                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     167 <b>Er</b> Erbium 68                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     169 <b>Tm</b> Thulium 69                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     173 <b>Yb</b> Ytterbium 70                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     175 <b>Lu</b> Lutetium 71                 </div>
<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     232 <b>Th</b> Thorium 90                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     234 <b>Pa</b> Protactinium 91                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     238 <b>U</b> Uranium 92                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     237 <b>Np</b> Neptunium 93                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     244 <b>Pu</b> Plutonium 94                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     247 <b>Am</b> Americium 95                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     251 <b>Cm</b> Curium 96                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     257 <b>Bk</b> Berkelium 97                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     261 <b>Cf</b> Californium 98                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     267 <b>Es</b> Einsteinium 99                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     271 <b>Fm</b> Fermium 100                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     277 <b>Md</b> Mendelevium 101                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     289 <b>No</b> Nobelium 102                 </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;">                     289 <b>Lr</b> Lawrencium 103                 </div>

The volume of one mole of any gas is 24 dm<sup>3</sup> at room temperature and pressure (r.t.p.).