



UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS
International General Certificate of Secondary Education

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CO-ORDINATED SCIENCES

0654/13

Paper 1 Multiple Choice

May/June 2012

45 minutes

Additional Materials: Multiple Choice Answer Sheet
Soft clean eraser
Soft pencil (type B or HB is recommended)

* 5 1 2 8 3 2 6 4 6 1 *

READ THESE INSTRUCTIONS FIRST

Write in soft pencil.

Do not use staples, paper clips, highlighters, glue or correction fluid.

Write your name, Centre number and candidate number on the Answer Sheet in the spaces provided unless this has been done for you.

There are **forty** questions on this paper. Answer **all** questions. For each question there are four possible answers **A, B, C** and **D**.

Choose the **one** you consider correct and record your choice in **soft pencil** on the separate Answer Sheet.

Read the instructions on the Answer Sheet very carefully.

Each correct answer will score one mark. A mark will not be deducted for a wrong answer.

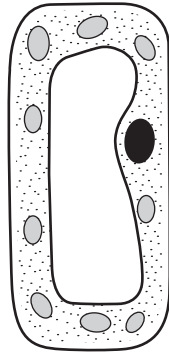
Any rough working should be done in this booklet.

A copy of the Periodic Table is printed on page 20.

This document consists of **17** printed pages and **3** blank pages.

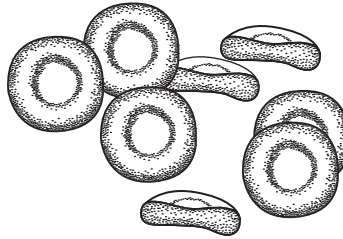


- 1 The diagram shows an incomplete plant cell.



Which structure is **not** shown?

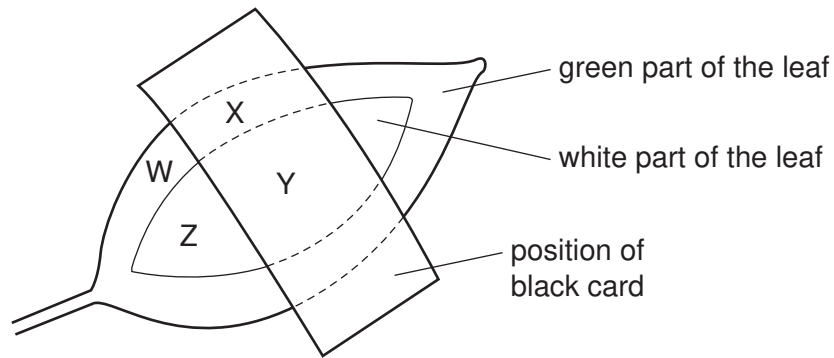
- A cell membrane
 - B cell wall
 - C chloroplast
 - D vacuole
- 2 The diagram shows some red blood cells.



What is the function of these cells?

- A antibody formation
- B carry oxygen
- C make glucose
- D phagocytosis

- 3 A plant has leaves which are green and white. One of the leaves is partly covered by a piece of black card. The plant is left under a lamp for two days.



The leaf is then removed from the plant and tested for starch (the carbohydrate made during photosynthesis).

Which row is correct?

	W	X	Y	Z
A	✓	✓	x	x
B	✓	x	x	x
C	x	✓	✓	x
D	x	x	✓	✓

key

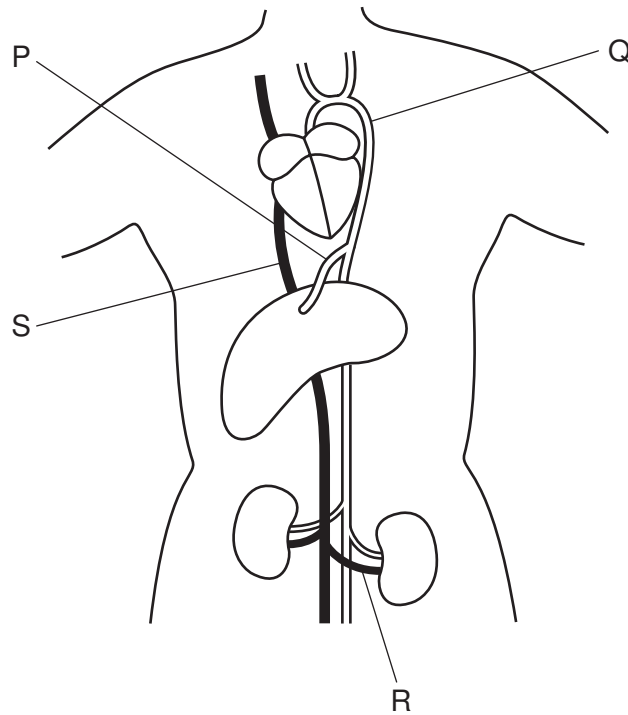
✓ = starch present

x = starch absent

- 4 Which word equation represents aerobic respiration?

- A** carbon dioxide + oxygen → glucose + water
B carbon dioxide + water → glucose + oxygen
C glucose + oxygen → carbon dioxide + water
D glucose + oxygen → lactic acid

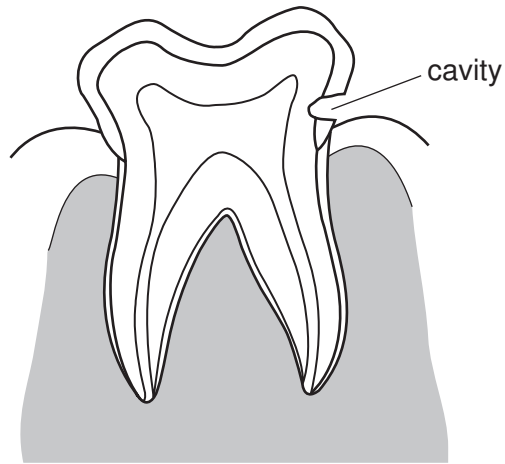
- 5 The diagram shows the heart, liver and kidneys with connecting blood vessels.



What are the labelled blood vessels?

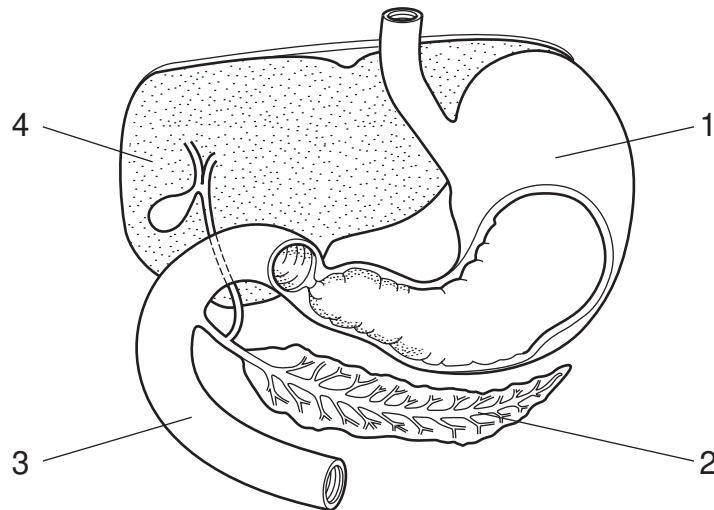
	aorta	hepatic artery	vena cava	renal vein
A	Q	P	S	R
B	Q	R	S	P
C	S	P	Q	R
D	S	R	Q	P

- 6 The diagram shows a tooth with a cavity caused by decay.



Which part of the tooth has been exposed by the decay?

- A enamel
 - B dentine
 - C pulp
 - D root
- 7 The diagram shows part of the digestive system.



Which labels identify the liver, pancreas and stomach?

	liver	pancreas	stomach
A	2	1	3
B	2	3	1
C	4	2	3
D	4	2	1

8 A person touches a hot object which triggers a reflex action.

In which order does the signal travel in the reflex arc?

- A relay neurone → spinal cord → sensory neurone
- B sensory neurone → spinal cord → motor neurone
- C spinal cord → sensory neurone → stimulus
- D stimulus → motor neurone → spinal cord

9 Which organ in the body makes urea and how is urea removed from that organ?

	where made	how removed
A	kidneys	in urine
B	liver	in blood
C	liver	in bile
D	pancreas	in pancreatic juice

10 Which structure produces seminal fluid?

- A prostate gland
- B sperm duct
- C testis
- D urethra

11 In plants, the allele for red flowers is dominant to the allele for white flowers.

A heterozygous red-flowered plant is crossed with a white-flowered plant.

What percentage of the offspring will have red flowers?

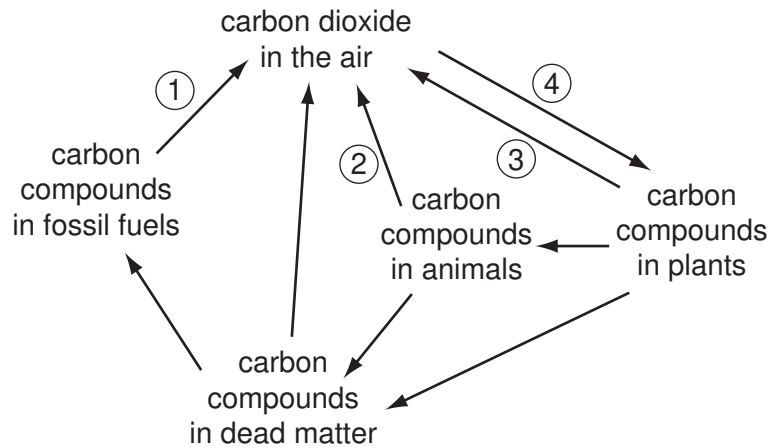
- A 25%
- B 50%
- C 75%
- D 100%

12 The diagram shows a food chain.

Which organisms pass the greatest amount of energy along the food chain?



13 The diagram shows part of the carbon cycle.

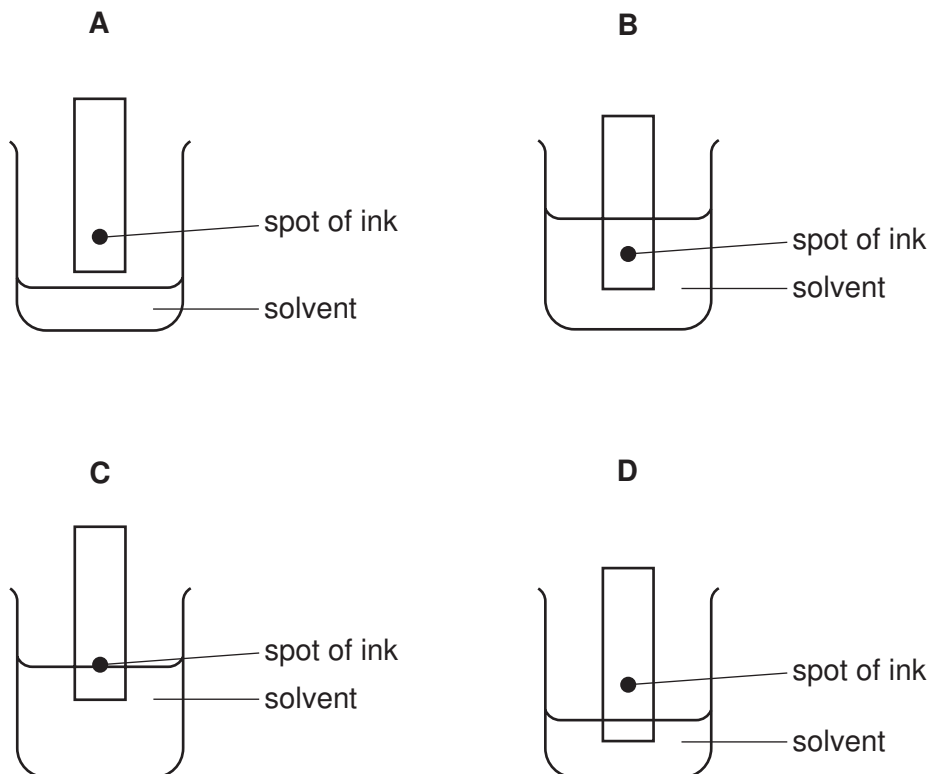


Where are photosynthesis and respiration happening?

	photosynthesis	respiration
A	3	2
B	3	4
C	4	1
D	4	3

14 The colours in an ink can be separated by chromatography.

Which diagram shows the correct way to set up the apparatus?



15 The symbol for an atom of neon is ${}_{10}^{20}\text{Ne}$.

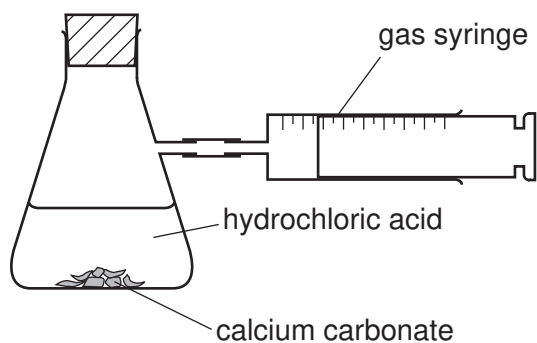
Which statement about the atom is correct?

- A It contains half as many neutrons as protons.
- B It contains twice as many neutrons as protons.
- C The number of neutrons equals the number of protons.
- D The total number of neutrons and protons is thirty.

16 Which formula contains the most elements?

- A NaOH B Rb_2S C SiCl_4 D SnO_2

17 The apparatus shown is used to investigate the speed of reaction between hydrochloric acid and calcium carbonate.



The time to collect 50 cm^3 of gas is measured.

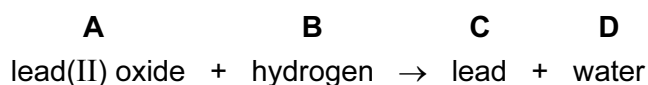
Using concentrated acid and lumps of calcium carbonate, the time is 150 s.

In a second experiment, the time is 90 s.

Which change was made in the second experiment?

- A larger lumps of calcium carbonate
- B less concentrated acid
- C lower temperature
- D powdered calcium carbonate

18 Which substance is reduced during the following reaction?

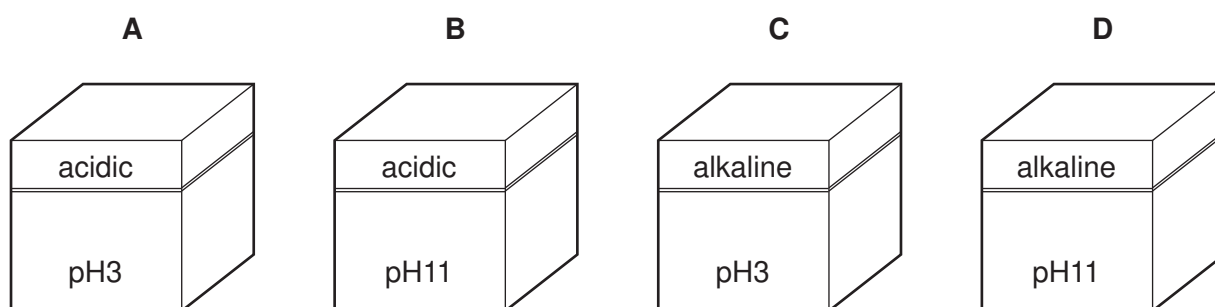


19 A coloured solution of compound X gives a white precipitate with acidified silver nitrate.

What is X?

- A calcium chloride
- B calcium sulfate
- C copper(II) chloride
- D copper(II) sulfate

20 Which box contains a solution of a metal oxide?



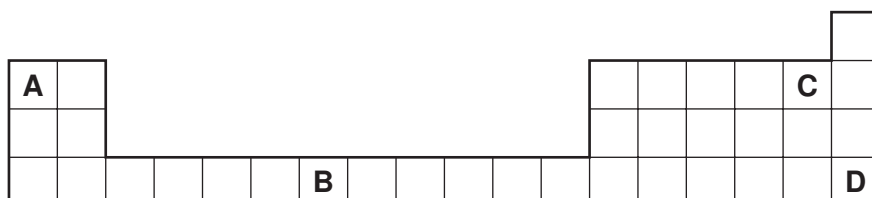
21 The table shows physical properties of some substances.

Which substance is a metal?

	malleability	density	electrical conductivity
A	brittle	high density	high
B	brittle	low density	low
C	malleable	high density	high
D	malleable	low density	low

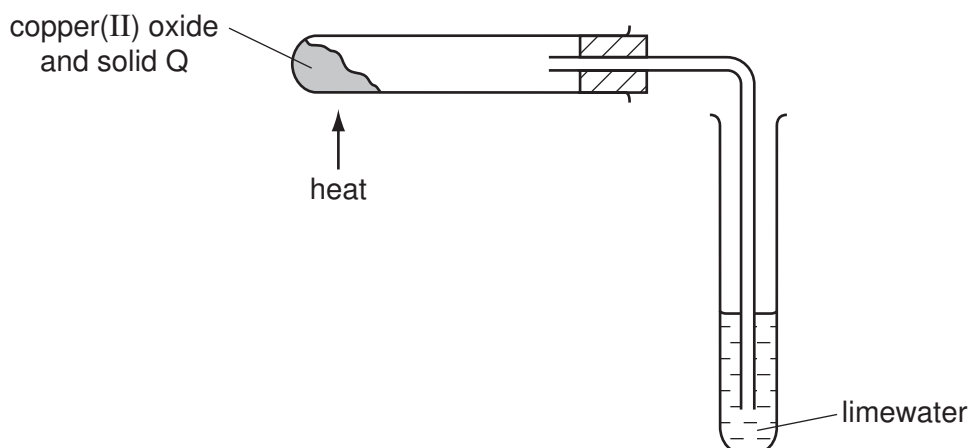
22 The diagram shows part of the Periodic Table.

Which letter shows the position of a metal with a low melting point?



23 Copper(II) oxide is mixed with a solid Q.

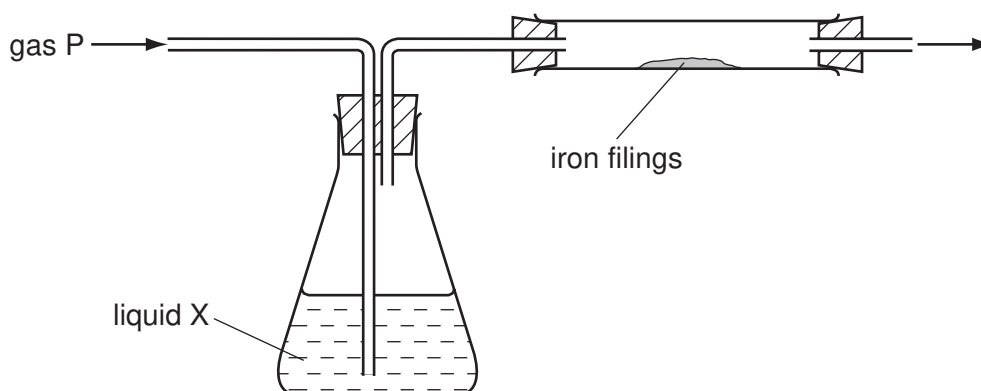
On heating the mixture, a reaction occurs and the limewater turns milky.



What is solid Q?

- A carbon
- B iron
- C sulfur
- D zinc

24 The diagram shows gas P being passed through liquid X and then over iron filings.



Which gas and liquid cause the iron to rust?

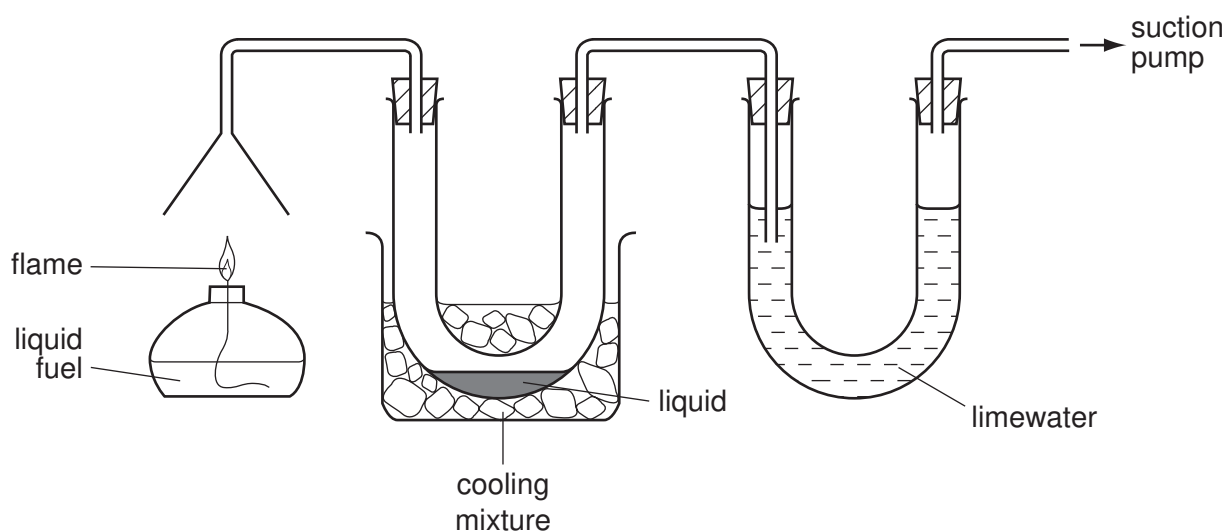
	gas P	liquid X
A	nitrogen	concentrated sulfuric acid (a drying agent)
B	nitrogen	water
C	oxygen	concentrated sulfuric acid (a drying agent)
D	oxygen	water

25 Fertilisers are used to supply the essential elements needed for plant growth.

Which compound supplies two of these essential elements?

- A $\text{Ca}(\text{H}_2\text{PO}_4)_2$
- B $\text{Ca}(\text{NO}_3)_2$
- C KNO_3
- D $(\text{NH}_4)_2\text{SO}_4$

26 The burning of a fuel is investigated using the apparatus shown.



Which substances is the apparatus testing for?

- A carbon monoxide and carbon dioxide
- B carbon monoxide and water
- C carbon dioxide and water
- D carbon dioxide and sulfur dioxide

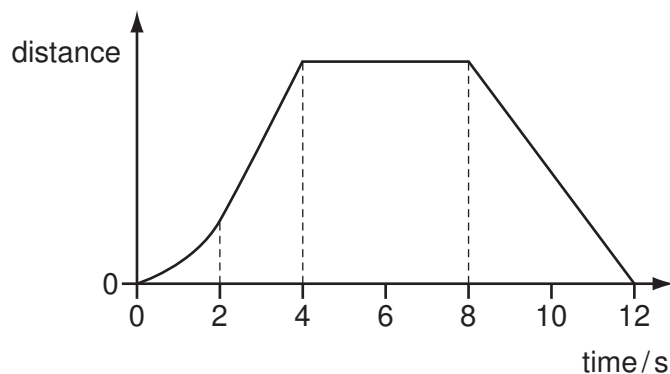
27 This description of a plastic is incomplete.

To make a plastic,1..... of a2..... combine to form a long chain3..... .

Which words correctly complete the gaps?

	1	2	3
A	atoms	monomer	polymer
B	atoms	polymer	monomer
C	molecules	monomer	polymer
D	molecules	polymer	monomer

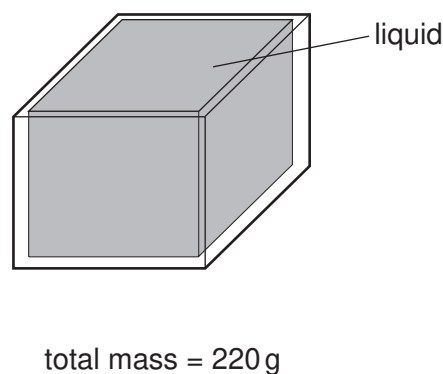
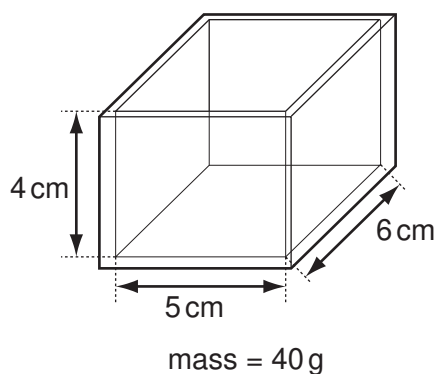
28 The graph shows how the distance of an object changes with time.



Between which two times is the object moving with a changing speed?

- A between 0 s and 2 s
- B between 2 s and 4 s
- C between 4 s and 8 s
- D between 8 s and 12 s

29 The diagrams show a glass tank with inside measurements of 5 cm × 6 cm × 4 cm.



The tank has a mass of 40 g when empty. When the tank is filled with a liquid, the tank and liquid have a total mass of 220 g.

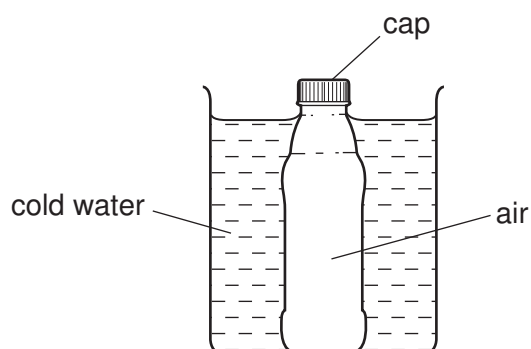
What is the density of the liquid?

- A $\frac{220}{(5 \times 6 \times 4)} \text{ g/cm}^3$
- B $\frac{(220 - 40)}{(5 \times 6 \times 4)} \text{ g/cm}^3$
- C $\frac{(5 \times 6 \times 4)}{220} \text{ g/cm}^3$
- D $\frac{(5 \times 6 \times 4)}{(220 - 40)} \text{ g/cm}^3$

30 Which energy resource did **not** receive its energy from the Sun?

- A geothermal
- B hydroelectric
- C oil
- D waves

31 A glass bottle containing air is sealed with a screw cap and then cooled in cold water.



The contraction of the glass bottle can be ignored.

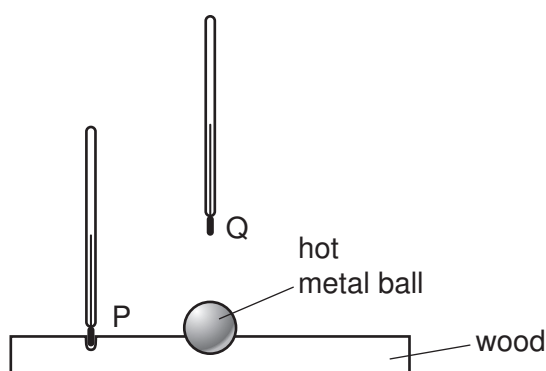
What remains the same during the cooling?

- A the air pressure inside the bottle
- B the energy of the air molecules in the bottle
- C the force on the cap made by the air molecules in the bottle
- D the volume of air inside the bottle

32 What happens to the temperature of a substance as it is melting and as it is boiling?

	melting	boiling
A	decreases	increases
B	decreases	no change
C	increases	increases
D	no change	no change

- 33 A hot metal ball is placed in a small hollow in a piece of wood. Two thermometers are placed at equal distances from the ball, one at position P and one at position Q.

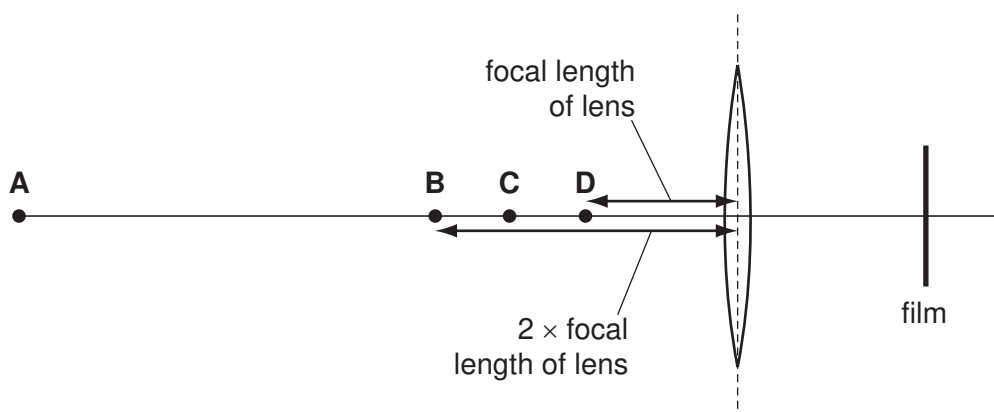


Which thermometer gives the highest reading and why?

	highest reading	reason
A	thermometer at P	the air conducts heat sideways, not upwards
B	thermometer at P	the wood conducts the heat sideways, not upwards
C	thermometer at Q	convection carries the heat upwards, not sideways
D	thermometer at Q	infra-red rays carry the heat upwards more than sideways

- 34 A converging lens in a camera is used to make an image on a film. The image is smaller than the object.

At which point could the object be placed so that it makes this image?

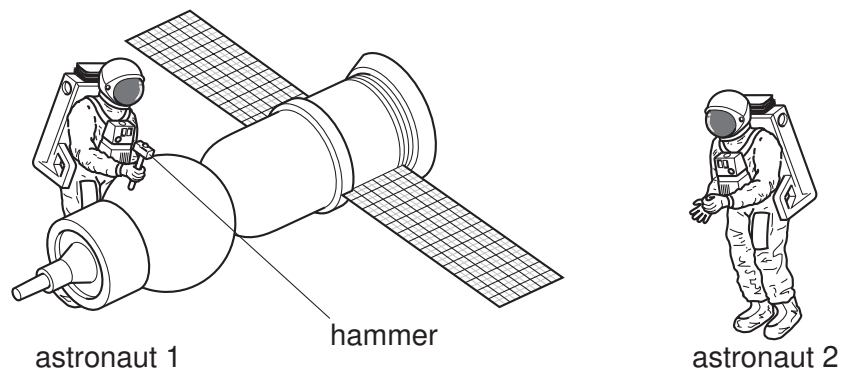


- 35 The Sun emits infra-red radiation, ultraviolet radiation and visible light.

Which statement about the time it takes these radiations to reach Earth's atmosphere is correct?

- A** Infra-red radiation arrives first.
- B** Ultraviolet radiation arrives first.
- C** Visible light arrives first.
- D** They all arrive at the same time.

- 36 Astronaut 1 uses a hammer to mend a satellite in space. Astronaut 2 is nearby. They are in space.



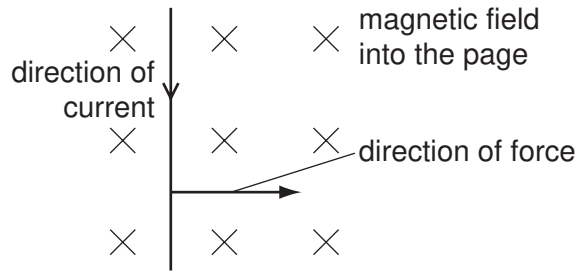
Compared with the sound heard if they were working on Earth, what does astronaut 2 hear?

- A a louder sound
 - B a quieter sound
 - C a sound of the same loudness
 - D no sound at all
- 37 The instructions for a household lamp state that the plug should be fitted with a 3 A fuse.

What could happen if, by mistake, a 13 A fuse is fitted?

- A The fuse might melt too easily.
- B The lamp might explode if a fault develops.
- C The wires connecting the lamp to the plug might overheat if a fault developed.
- D Too much voltage might be supplied to the lamp.

- 38 An electric current flows along a wire. A magnetic field is at right angles to it (into the page) causes a force on the wire to the right.

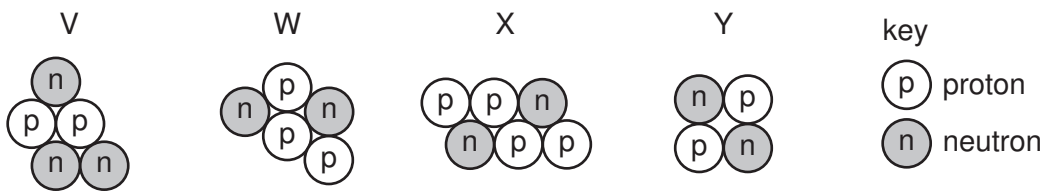


In which direction is the force when the current direction is reversed?

- A to the right
 - B to the left
 - C upwards
 - D downwards
- 39 Which row describes the nature of alpha and beta radiation?

	alpha	beta
A	electron	electromagnetic wave
B	electron	helium nucleus
C	helium nucleus	electromagnetic wave
D	helium nucleus	electron

- 40 The diagrams represent the nuclei of four different atoms V, W, X and Y.



Which two diagrams represent isotopes of the same element?

- A V and W
- B W and X
- C X and Y
- D Y and V

DATA SHEET

The Periodic Table of the Elements

Group																					
I	II											III	IV	V	VI	VII	0				
										1 H Hydrogen 1											4 He Helium 2
7 Li Lithium 3	9 Be Beryllium 4											11 B Boron 5	12 C Carbon 6	14 N Nitrogen 7	16 O Oxygen 8	19 F Fluorine 9	20 Ne Neon 10				
23 Na Sodium 11	24 Mg Magnesium 12											27 Al Aluminium 13	28 Si Silicon 14	31 P Phosphorus 15	32 S Sulfur 16	35.5 Cl Chlorine 17	40 Ar Argon 18				
39 K Potassium 19	40 Ca Calcium 20	45 Sc Scandium 21	48 Ti Titanium 22	51 V Vanadium 23	52 Cr Chromium 24	55 Mn Manganese 25	56 Fe Iron 26	59 Co Cobalt 27	59 Ni Nickel 28	64 Cu Copper 29	65 Zn Zinc 30	70 Ga Gallium 31	73 Ge Germanium 32	75 As Arsenic 33	79 Se Selenium 34	80 Br Bromine 35	84 Kr Krypton 36				
85 Rb Rubidium 37	88 Sr Strontium 38	89 Y Yttrium 39	91 Zr Zirconium 40	93 Nb Niobium 41	96 Mo Molybdenum 42	96 Tc Technetium 43	101 Ru Ruthenium 44	103 Rh Rhodium 45	106 Pd Palladium 46	108 Ag Silver 47	112 Cd Cadmium 48	115 In Indium 49	119 Sn Tin 50	122 Sb Antimony 51	128 Te Tellurium 52	127 I Iodine 53	131 Xe Xenon 54				
133 Cs Caesium 55	137 Ba Barium 56	139 La Lanthanum 57 *	178 Hf Hafnium 72	181 Ta Tantalum 73	184 W Tungsten 74	186 Re Rhenium 75	190 Os Osmium 76	192 Ir Iridium 77	195 Pt Platinum 78	197 Au Gold 79	201 Hg Mercury 80	204 Tl Thallium 81	207 Pb Lead 82	209 Bi Bismuth 83	210 Po Polonium 84	210 At Astatine 85	210 Rn Radon 86				
87 Fr Francium	226 Ra Radium	227 Ac Actinium †																			

*58-71 Lanthanoid series

†90-103 Actinoid series

Key

a	a = relative atomic mass
X	X = atomic symbol
b	b = proton (atomic) number

140 Ce Cerium 58	141 Pr Praseodymium 59	144 Nd Neodymium 60	147 Pm Promethium 61	150 Sm Samarium 62	152 Eu Europium 63	157 Gd Gadolinium 64	159 Tb Terbium 65	162 Dy Dysprosium 66	165 Ho Holmium 67	167 Er Erbium 68	169 Tm Thulium 69	173 Yb Ytterbium 70	175 Lu Lutetium 71
232 Th Thorium 90	234 Pa Protactinium 91	238 U Uranium 92	237 Np Neptunium 93	244 Pu Plutonium 94	247 Am Americium 95	247 Cm Curium 96	247 Bk Berkelium 97	251 Cf Californium 98	252 Es Einsteinium 99	257 Fm Fermium 100	261 Md Mendelevium 101	265 No Nobelium 102	269 Lr Lawrencium 103

The volume of one mole of any gas is 24 dm³ at room temperature and pressure (r.t.p.).

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