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UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS

International General Certificate of Secondary Education

MARK SCHEME for the June 2005 question paper

0654 CO-ORDINATED SCIENCES

0654/02

Paper 2 Core (Theory), maximum raw mark 100

Mark schemes must be read in conjunction with the question papers and the Report on the Examination.

CIE will not enter into discussion or correspondence in connection with these mark schemes.

CIE is publishing the mark schemes for the June 2005 question papers for most IGCSE and GCE Advanced Level and Advanced Subsidiary Level syllabuses and some Ordinary Level syllabuses.

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Grade thresholds for Syllabus 0654 (Co-ordinated Sciences) in the June 2005 examination.

	maximum	minimum mark required for grade:				
	mark available	AA	CC	EE	FF	
Component 2	100	N/A	50	32	24	

The threshold (minimum mark) for DD is set halfway between those for Grades CC and EE. The threshold (minimum mark) for GG is set as many marks below the FF threshold as the EE threshold is above it.

June 2005

IGCSE

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MARK SCHEME

MAXIMUM MARK: 100

SYLLABUS/COMPONENT: 0654/02

CO-ORDINATED SCIENCES
Paper 1 Core (Theory)

	Page 1		Mark Scheme Syllabus	10
			IGCSE – June 2005 0654	200
1	(a)	(i) (ii)	hydrogen; R (most) Q S (least);	7. PanaCo [1]
		(iii)	increase in temperature of reactants/tube feels warm/owtte	; [1]
	(b)	(i) (ii)	water; two electrodes dipping into electrolyte; electrodes connected across voltmeter;	[1] [2]
		(iii)	reacts with water/it will dissolve/is very reactive;	[1]
2	(a)	(i)	as force increases length of wool fibre increases; proportionally/owtte;	[2]
		(ii)	breaks/will not return to normal length/loses elasticity/ past elastic limit	[1]
	(b)		traps air; acts as insulator; less convection/less heat loss by radiation;	[max 2]
	(c)		genes; environment; or two environmental factors e.g. temperature/climate; food/diet;	[2]
	(d)	(i) (ii)	scaly; no covering/smooth/moisture;	[2]
}	(a)	(i)	B no mark mass/weight ratio = 10/owtte;	[1]
		(ii)	A and D;	[1]
		(iii)	C no mark has least mass;	[1]
	(b)		particles; vibrate;	[2]
	(c)		time = distance/speed; = 400 000/300 000 = 1.33 seconds;	[2]
4	(a)		violet and green;	[1]
	(b)		would not contain manganese oxide; would not contain iron oxide; reference to avoiding transition metal compounds/ transition metal compounds impart the colour;	[3]
	(c)		B; reference to giant structure or description/B shows a compo reference to random arrangement of atoms;	ound; [3]

		1			1 2 11 1	20
	Page 2		la la	Mark Scheme GCSE – June 2005	Syllabus 0654	(B)
	(d)	(i) (ii)	oxygen; sulphur diox sulphur diox	ide is formed; ide harmful to humans; ulphur dioxide must be kept		[max: 2]
5	(a)		A stamen/ar B ovary/ovul C petal;	The state of the s		[3]
	(b)			3;		[max: 3]
	(c)	(i) (ii)	energy trans	light; d to make carbon dioxide a	nd water react;	[1] [max: 2]
	(d)	(i) (ii)	organisms ir	→ bee →bee eater n correct order; n pointing the right way; e eaters;		[2] [1]
6	(a)	(i)				
			Ivantage sadvantage	no pollution/ saves fossil fuels/ renewable need lots of turbines/ damages landscapes/ noisy/ no use without wind	no air pollution/ one nuclear power station provides lots of energy waste needs safe disposal/ radioactive waste harms people/ waste needs safe disposal	
		(ii)	fossil fuels a	re running out/causing pollu		;;;; [4] [1]
	(b)	(i) (ii)	transformers to reduce he	s; eat/energy losses;		[1] [1]
7	(a)			icken and fat to butter; aking new cells and fat to pr	roviding	[2]
	(b)	(i) (ii)	liver; by kidneys; detail of how (excreted in)	they work e.g. filtration from urine;	m blood;	[1] [2]

	Page 3		Mark Scheme	Syllabus	1
	rage 3		IGCSE – June 2005	0654	00
8	(a)		light can pass through/you can see through	Syllabus 0654	Cal
	(b)		contain the same elements/covalent; different numbers of atoms one more atom of the peroxide/peroxide molecule larger/heavier	oxygen in	[2]
	(c)	(i)	glowing splint; relights;		[2]
		(ii)	1.0g/the same; catalyst is not used up/owtte;		[2]
	(d)		add detergent to water; use a non-aqueous solvent;		[2]
9	(a)	(i) (ii)	not a complete circuit/no connection between ampere;	cell and lamp;	[1] [1]
	(b)	(i)	ammeter in series; voltmeter in parallel; rest of circuit correct;		[3]
		(ii)	R = V/I; evidence of working; = 1 ohm;		
		(iii) (iv)	0.4 x 0.4 = 0.16; 0.16;		[3] [1] [1]
10	(a)		cannot be created nor destroyed; always converted into another form;		[2]
	(b)		energy initially used to make the particles mov at boiling point energy used to separate particl forces of attraction		[2]
	(c)		water is a good conductor of electricity; danger of electric shock/electrocution;		[2]
	(d)		smaller volume; means more collisions with walls of container; more collisions with wall means more pressure	e;	[2]
11	(a)		electrolyte; anode;		[2]
	(b)		solution becomes alkaline;		[2]

sodium hydroxide produced;

(i) (ii) (iii) (iv)

five; five;

seven; neutron;

(c)

[2]

[1] [1] [1] [1]

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ı	Page 4		Mark Scheme	Syllabus	1.0
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12	(a)	(i) (ii) (iii)	protects the brain; cartilage softer; any named synovial joint e.g. elbow, knee; reduced friction/provides smooth surface; or in nose/ear; support with flexibility;		Papacambridge com
	(b)	(i) (ii)	chewing/grinding; increase surface area of food; food more likely to stay on them;		[2]
			because they have uneven surface; bacteria feed on the food/bacteria produce acids	;;	[max: 2]