UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS International General Certificate of Secondary Education

## www.papacambridge.com MARK SCHEME for the May/June 2008 guestion paper

## 0654 CO-ORDINATED SCIENCES

0654/02

Paper 2 (Core Theory), maximum raw mark 100

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began.

All Examiners are instructed that alternative correct answers and unexpected approaches in candidates' scripts must be given marks that fairly reflect the relevant knowledge and skills demonstrated.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

CIE will not enter into discussions or correspondence in connection with these mark schemes.

CIE is publishing the mark schemes for the May/June 2008 question papers for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level syllabuses and some Ordinary Level syllabuses.

1 4	ge 2		Syllabus er
		IGCSE – May/June 2008	0654 73
(a)		nea, lens; ore pupil, humours)	Syllabus 0654 PhoCambridge Office Off
(b)	(i)	focuses/adjusts light/image; onto the retina; lens changes shape; ref to refraction/bending light;	[max 2]
	(ii)	contains receptor/light sensitive cells; converts light energy to impulse in nerve (fibre); impulse sent to brain;	[max 2]
(c)	(i)	abnormal choroid/blindness;	[1]
	(ii)	gametes <b>A</b> and <b>a</b> ; offspring <b>AA</b> and <b>Aa</b> ; all normal/none have disease; (allow ecf)	[3] [Total: 9]
(a)	den = 4(	sity = mass/volume; ) / 35 = 1.14 g / cm <sup>3</sup> ;	[2]
(b)	= 0.	mentum = mass x velocity; 04 x 40 6 kg m/s;	[2]
(c)	(i)	60 N;	[1]
	(ii)	work = force x distance; = 60 x 0.5	
		= 30 J; (allow ecf)	[2]

Page 3	3 Mark Scheme	Syllabus
	IGCSE – May/June 2008	0654
<b>(a) A</b> /iq	gneous;	Coll.
( )		Otio
(b) (i)	sedimentary;	Syllabus 0654 Photoscontinution Notest
(ii)	(biological)	
• •	roots;	
	abrade rock surface; animals;	
	ahmais, abrade rock surface;	
	(physical)	
	description of freeze/thaw;	
	reference to ice expansion;	
	description of thermal variation; expansion/contraction cause surface damage;	
	particles carried by wind;	
	abrade rock surface;	
	(chemical)	
	(acidic) rain;	
	reacts with rock/dissolves rock;	[max 2]
(iii)	correct underlined from (ii)	[1]
(c) (i)	colloid;	[1]
(ii)	(incorrect)	
	should be called a sol;	[0]
	emulsion is liquid in liquid / sol is name for solid in liquid;	[2]
(iii)	water contains (dissolved) sulphate (ions);	[1]
		[Total: 9]
(a) (i)	A = palisade (layer);	
	<b>B</b> = (lower) epidermis;	[2]
(ii)		
	it has chloroplasts/chlorophyll;	
	it has a vacuole/cell sap; it can photosynthesise;	[max 2]
(iii)	arrow drawn entering stoma;	[1]
(b) car	rries water (to the leaf);	
car	rries minerals;	[mov 0]
sup	oport;	[max 2]

0	Syllabus	Mark Se	ge 4	Pa
Dec	0654	IGCSE – May		<u> </u>
ants.		<b>S</b> on a horizontal portion;	(i)	(a)
Daba Cambridas [1]		goes faster/accelerates/accele	(ii)	
[1]		number of waves (produced) p	(i)	(b)
[1]		dolphin;	(ii)	
[1]		dolphin;	(iii)	(
[3]		tance = speed x time; 500 x 0.2 = 300m; tance = 150m ;	= 15	(c)
	straight lines with arrows; bending at surface;		(d)	
[3]		entering eye;		
[Total: 11]				
[max 2]		e.g. lithium is less dense; has higher melting point; is less malleable; is less reactive;		(a)
[1]		electron configuration 2,8 show	(ii)	
[1]	proton than electron;	ions form by losing one electro	(iii)	(
[2]	ing ions;	magnesium sulphate; both soluble and ionic/electroly		(b)
[max 1]	lectrodes;	use different metals/materials	• •	
[Total: 7]				
[1]		May;	(i)	(a)
[2]		idea that it was lower (except i idea that it peaked at different		
[max 2]		plants use nitrate to make prot plants grow, larger/better/faste higher yield/bigger crop;		(b)
[1]		add (nitrogen-containing) fertili	(ii)	

	Mark Scheme	Syllabus Syllabus
	IGCSE – May/June 2008	0654 230
(c) (i) maiz	ze	and,
(ii) enei	rgy (flow);	Syllabus 0654 Pharcannbrigg
• •	osers/named decomposer; pots/break them down/decomposes;	
	on (by composers) releases carbon dioxide;	[max 2]
		[Total: 10]
	nal bodywork attracted;	101
	d hole not attracted;	[2]
.,	tic filler is not magnetic	[1]
. ,	- aluminium is not magnetic;	[1]
(iv) alun	ninium doesn't corrode/corrodes less than steel/less	dense; [1]
<b>(b)</b> In a	SOLID , the particles are closer together t	han in a <u>GAS</u> .
The	e forces of attraction between particles are stronger i	in a <u>SOLID</u> than in a <u>GAS</u>
Wh	en a SOLID is heated it will eventually turr	n into a liquid.
In a	SOLID, the particles can only vibrate and no	ot move.
Hea	at energy will travel through a <u>SOLID</u> by conc	duction.
Неа	at energy will <b>not</b> travel through a <u>SOLID</u> by o	convection.
Any two	correct 1 mark	[4]
		[Total: 9]
(a) made fro	om once living material/millions of years to form;	[1]
	lioxide produced;	
	e to (excessive) global warming/enhanced greenhou e to negative consequences of climate change;	use effect; [max 2]
. , . ,	ewater; s cloudy;	נסז
-		[2]
.,	her % of methane/more methane; hane burns/other gases do not burn/contribute to he	eat output; [2]
(a) speeds u		[Total: 7]



