

## **MARK SCHEME for the May/June 2014 series**

### **0620 CHEMISTRY**

**0620/61**

Paper 6 (Alternative to Practical), maximum raw mark 60

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge will not enter into discussions about these mark schemes.

Cambridge is publishing the mark schemes for the May/June 2014 series for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level components and some Ordinary Level components.

Page 2	Mark Scheme	Syllabus	Paper
	IGCSE – May/June 2014	0620	61

- 1 (a) thermometer (1)  
condenser (1)  
**allow** condensing tube, condensating tube, etc. [2]
- (b) arrows labelled – water (in) and water (out) (1) [1]
- (c) fractional (1)  
distillation (1) [2]
- (d) (i) ethanol (1) [1]  
(ii) temperature would rise (above 78°C) (1) [1]
- (e) alcohols are (in)flammable / catch fire / burn (1)  
**ignore:** explode  
Bunsen burner / flame / heat (1) [2]
- 2 (a) precipitation / double decomposition (1) [1]  
**allow:** ppt
- (b) (i) low / insoluble / does not dissolve (1) [1]  
(ii) high / soluble / dissolves (1) [1]
- (c) filtration (1) [1]
- 3 (a) volumes of hydrogen completed correctly (3)  
0, 8, 34, 42, 46, 48, 48 [3]  
**guidance:** 7 correct (3); 6 correct (2); 5 correct (1); 4 or fewer correct (0)
- (b) points plotted correctly including origin (3)  
**guidance:** 7 correct (3); 6 correct (2); 5 correct (1); 4 or fewer correct (0)  
smooth curve missing anomalous point (1) [4]

Page 3	Mark Scheme	Syllabus	Paper
	IGCSE – May/June 2014	0620	61

- (c) (i) point at  $5\text{ cm}^3 / 8\text{ cm}^3 \text{ H}_2$  / second point (1) [1]
- (ii) leak / loss / escape of gas or wrong amount / too little HC / or zinc (1) [1]  
**allow:** syringe sticking
- (iii) reading from graph (1)  $\pm$  half small square  
indication on graph (1) [2]
- (d) excess acid (1)  
all zinc reacted (1) [2]  
**allow:** used up
- (e) sketch curve identical (2)  
different curve levelling out at  $48\text{ cm}^3$  (1) [2]  
**note:** must be some indication of a second curve
- 4 (a) table of results for Experiment 1  
initial and final volume boxes completed correctly (1) 0.0 and 16.8  
difference box correctly completed (1) 16.8  
all readings to one decimal place (1) [3]
- (b) table of results for Experiment 2  
initial (1) and final volume (1) boxes completed correctly 16.8 (1) and 25.2 (1)  
difference box correctly completed (1) 8.4 [3]
- (d) to colourless (1) [1]  
**not:** clear
- (e) coloured reacting mixture masks colour of phenolphthalein / reaction is finished / solution is acidic (1) [1]
- (f) carbonate / carbon dioxide present (1) [1]  
**allow:** hydrogencarbonate

Page 4	Mark Scheme	Syllabus	Paper
	IGCSE – May/June 2014	0620	61

- (g) (i) 8.4 (1)  
**ecf:** titre 1 – titre 2  
 cm<sup>3</sup> (1) [2]
- (ii) 16.8 (1) [1]  
**ecf:** 2 × titre 2
- (iii) twice volume of acid needed to react with T (1) [1]  
**ecf:** if (g)(i) or / and (g)(ii) wrong need quantitative link.  
**not:** more (unqualified)

- (h) (i) 67.2 cm<sup>3</sup> (1)  
 33.6 cm<sup>3</sup> (1)  
 4 × volume of solution R (1) [3]
- (ii) volume of acid used > 50 cm<sup>3</sup> / more than burette can hold (1)  
 set up more than two burettes / 100.8 won't fit into 2 (1) [2]  
**allow:** impurities / contamination (1)

- 5 (d) white (1)  
 precipitate (1) [2]
- (e) no reaction / no change / no precipitate (1) [1]  
**allow:** colourless solution
- (f) not a chloride / halide (1) [1]
- (g) oxygen / O<sub>2</sub> (1) [1]  
 not O
- (h) transition metal / manganese (1)  
 hydrated salt (1)  
**ignore:** sulfate  
**allow:** catalyst (1) max [2]

<b>Page 5</b>	<b>Mark Scheme</b>	<b>Syllabus</b>	<b>Paper</b>
	<b>IGCSE – May/June 2014</b>	<b>0620</b>	<b>61</b>

- 6 same / measured volume of water (1)
- initial temperature (1)
- mass of nut(s) (1)
- ignite / burn (1)  
**not:** heat
- for suitable time < 10 minutes / to completion (1)
- final temperature of water (1)
- repeat with other nut(s) (1)
- compare / conclusion (1) max [7]