

## **MARK SCHEME for the May/June 2014 series**

### **0620 CHEMISTRY**

**0620/63**

Paper 6 (Alternative to Practical), maximum raw mark 60

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

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- 1 (a) (i) water (1) [1]  
**not:** steam
- (ii) two arrows, one under magnesium, one on wool (1) [1]
- (b) (i) grey/silver (1)  
white (ash) (1)  
glows/ignites/burns (1) max 2 [2]
- (ii) dissolves/forms solution/alkali (1)  
blue/purple/pH>7 (1) [2]
- (c) catches fire/explodes/pops (1)  
hydrogen (1) [2]
- 2 (a) labels on both graphs, i.e. Experiment 2 on that levelling at 60 and Experiment 1 on graph levelling at 30 (1) [1]
- (b) (i) water (1)  
25 cm<sup>3</sup> of dilute acid + 25 cm<sup>3</sup> of water/equal volumes (1) [2]
- (ii) graph less steep than others (1)  
levelling at 15 (1) [2]
- (c) gas syringe **or** measuring cylinder inverted in trough of water (1)  
labelled collection vessel/graduations shown on collection vessel (1) [2]
- (d) heat/increase temperature (1)  
particles have more energy/move faster (1)  
more frequent/more successful/more collisions(1)
- OR**
- catalyst (1)  
lowers activation energy (1)  
more successful collisions (1) [3]

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- 3 (a) electroplating (1) [1]  
**allow:** electrolysis
- (b) to clean/remove dirt/impurities (1)  
so nickel coats evenly/efficiently (1) [2]
- (c) aqueous/solution in water(1)  
named nickel salt (1) [2]  
**allow:** nickel ions
- (d) bulb lights/(silver) deposit on key (1) [1]
- (e) rinse with water and suitable method to dry e.g. oven/hairdryer (1) [1]
- 4 (a) temperature boxes correctly completed (2),  
25, 36, 38, 37, 36, 35, 34 [2]  
**guidance:** 7 correct (2); 6 correct (1); 5 or fewer correct (0)
- (b) temperature boxes completed correctly  
25, 19, 18, 17, 16, 16, 17 [2]  
**guidance:** 7 correct (2); 6 correct (1); 5 or fewer correct (0)
- (d) all points correctly plotted(3)  
**guidance:** 7 correct (2); 6 correct (1); 5 or fewer correct (0)  
smooth line graphs (2)  
labels (1) [6]
- (e) (i) value from graph (1) 37.5 s  
shown clearly (1) [2]
- (ii) value from graph (1) 6 s  
shown clearly (1) [2]
- (f) endothermic (1) [1]
- (g) M is a carbonate/carbon dioxide given off (1) [1]

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(h) lower temperature changes (1)  
greater volume / more water (1) [2]

(i) room temperature or 25 °C (1)  
reaction finished (1) [2]

(j) more readings / points / more accurate / better graph (1) [1]

5 (c) (i) white (1)  
precipitate(1)  
insoluble(1) [3]  
(ii) no / thin precipitate (1) [1]  
(iii) yellow precipitate (1) [1]

(d) copper (1)  
oxide (1) [2]

6 x cm<sup>3</sup> of vinegar (1)  
in named container e.g. beaker (1)  
add named indicator (1)  
add sodium hydroxide until colour change (1)  
record volume sodium hydroxide added (1)  
repeat with other vinegar (1)  
compare results (1) [7]