## UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS General Certificate of Education Advanced Subsidiary Level and Advanced Level

CHEMISTRY 9701/05

Paper 5 Practical Test

May/June 2005

CONFIDENTIAL INSTRUCTIONS

Great care should be taken that any confidential information given does not reach the candidates either directly or indirectly.

## **ERRATUM**

Page 4, Detailed Identities of Chemicals Required

Solutions, Question 1 and Question 2

For Centres using approximately 36% hydrochloric acid (11.7 M), the following instructions should be used to prepare solutions **FB 1** and **FB 4**:

**FB 1** is 2.0 mol dm<sup>-3</sup> hydrochloric acid, prepared by diluting 172 cm<sup>3</sup> of concentrated hydrochloric acid to 1 dm<sup>3</sup>.

**FB 4** is 2.2 mol dm<sup>-3</sup> hydrochloric acid, prepared by diluting 189 cm<sup>3</sup> of concentrated hydrochloric acid to 1 dm<sup>3</sup>.

Titrate 25 cm<sup>3</sup> portions of **FB 4** against **FB 3** and adjust the concentration of **FB 4** to give a titre of  $36.7 \pm 0.2$  cm<sup>3</sup>.